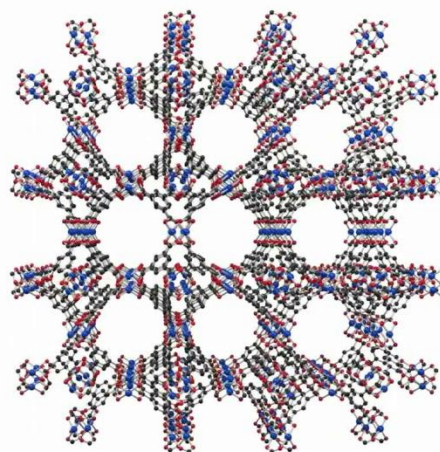
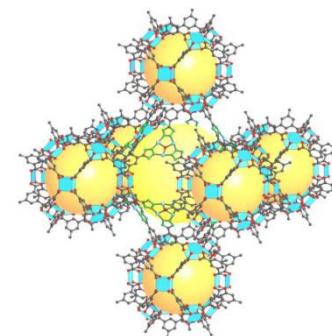
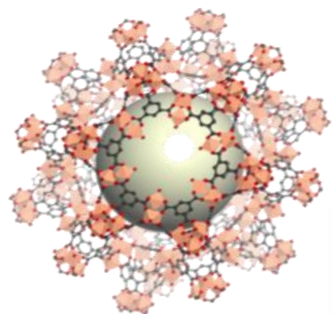




Reticular Chemistry: MOF Design Strategies to Applications

(Molecular Precision in infinite 2D and 3D)

Konstantinos Froudas (1039)



Supramolecular
Chemistry

Heraklion, 4/5/2020

Contents

Emergence of
MOFs

1

Design

2

Synthesis-
Activation of
MOFs

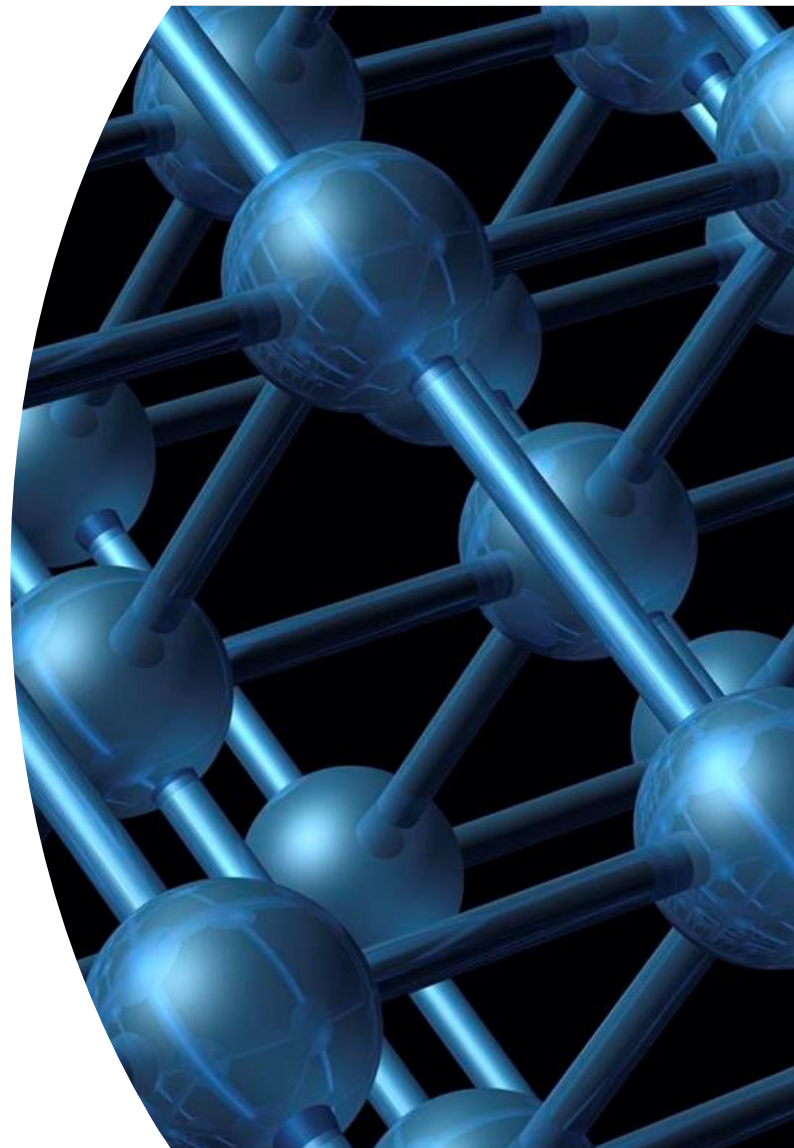
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Applications

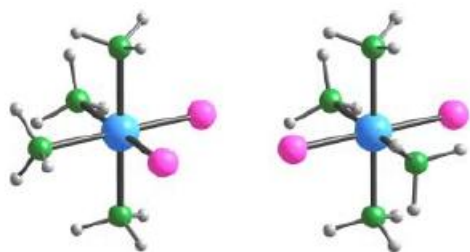
4

What's next?

5



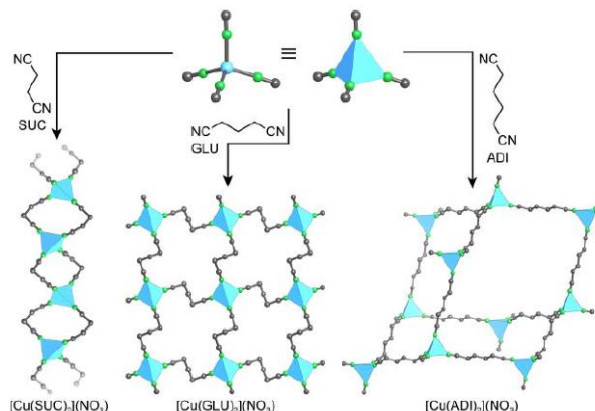
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Werner complexes
(Nobel Prize, 1993)

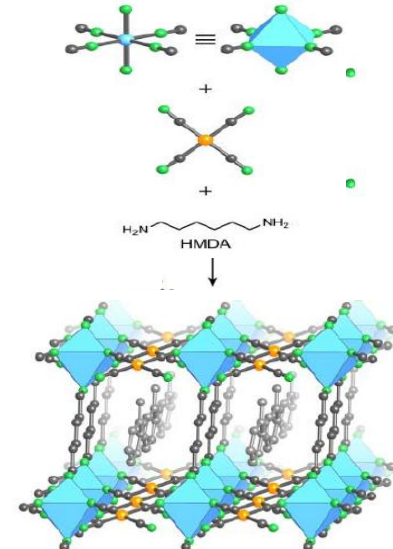
Original Formulation	Color	Ions per Formula Unit	"Free" Cl ⁻ Ions per Formula Unit	Modern Formulation
CoCl ₃ · 6 NH ₃	Orange	4	3	[Co(NH ₃) ₆]Cl ₃
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1893



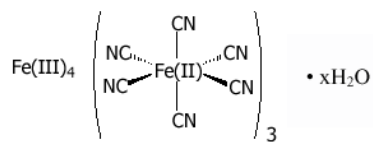
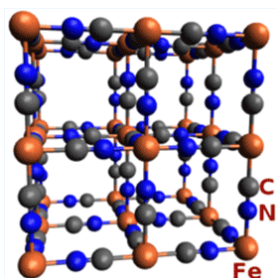
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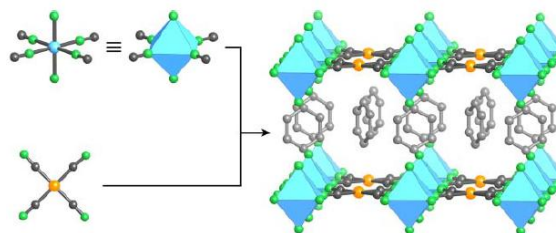
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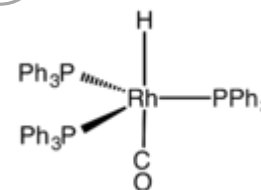
1897



Hofmann clathrates
Ni(CN)₂(NH₃)

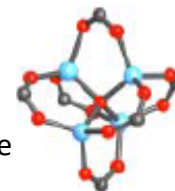


Development of the
Organometallic
Chemistry



1983-1984

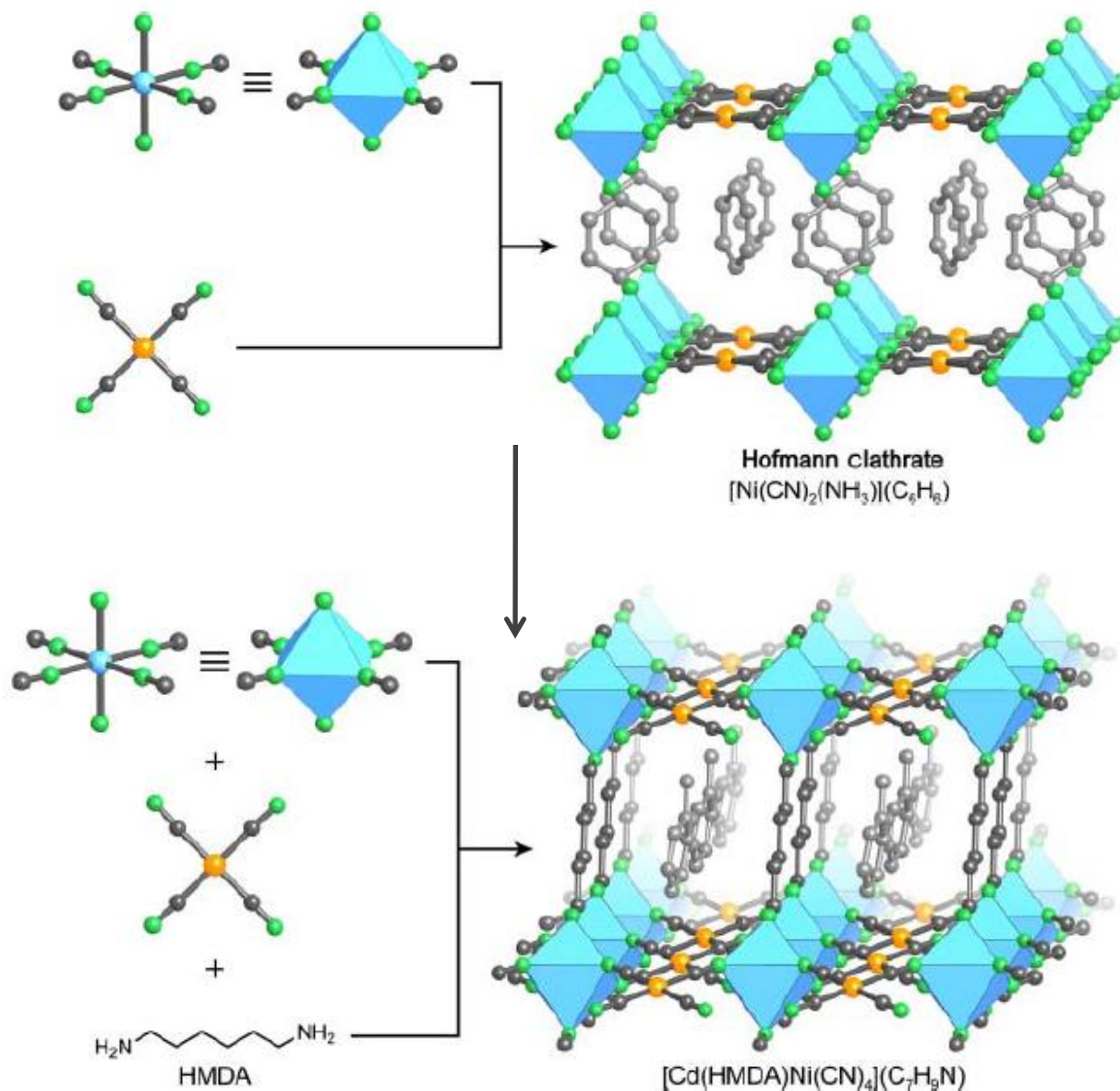
Basic Zinc Acetate



Extended structures
Molecular chemistry



Hofmann Clathrates

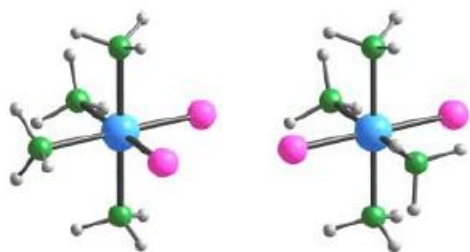


K. Hofmann and F. Kóspert, *Zeitschrift für Anorganische Chemie*, **1897**, 15 (1), 204–207.

S.-I. Nishikiori and T. Iwamoto, *Chemistry Letters*, **1984**, 13 (3): 319–322.



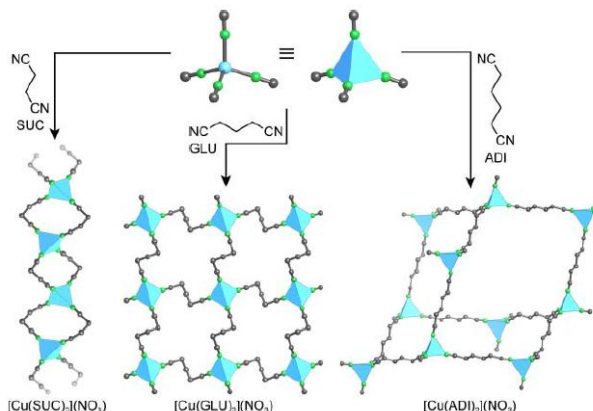
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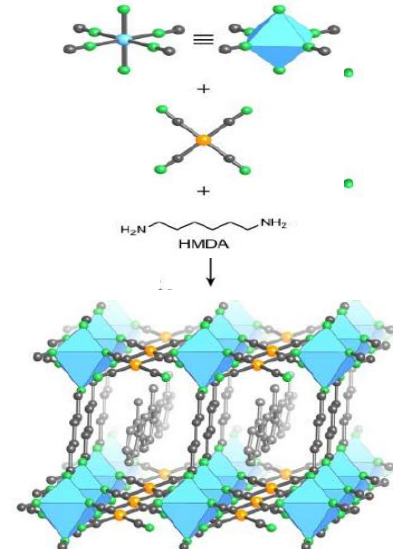
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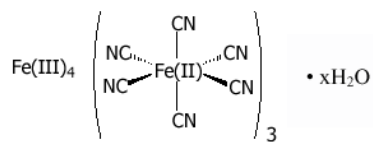
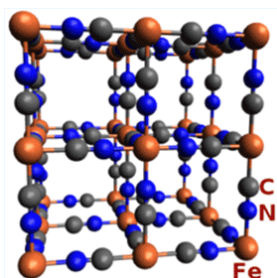
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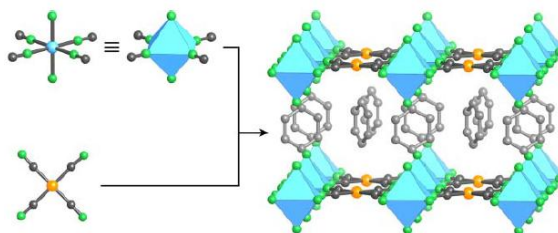
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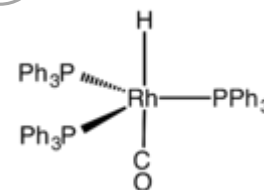
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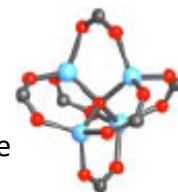


Development of the
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1983

Basic Zinc Acetate

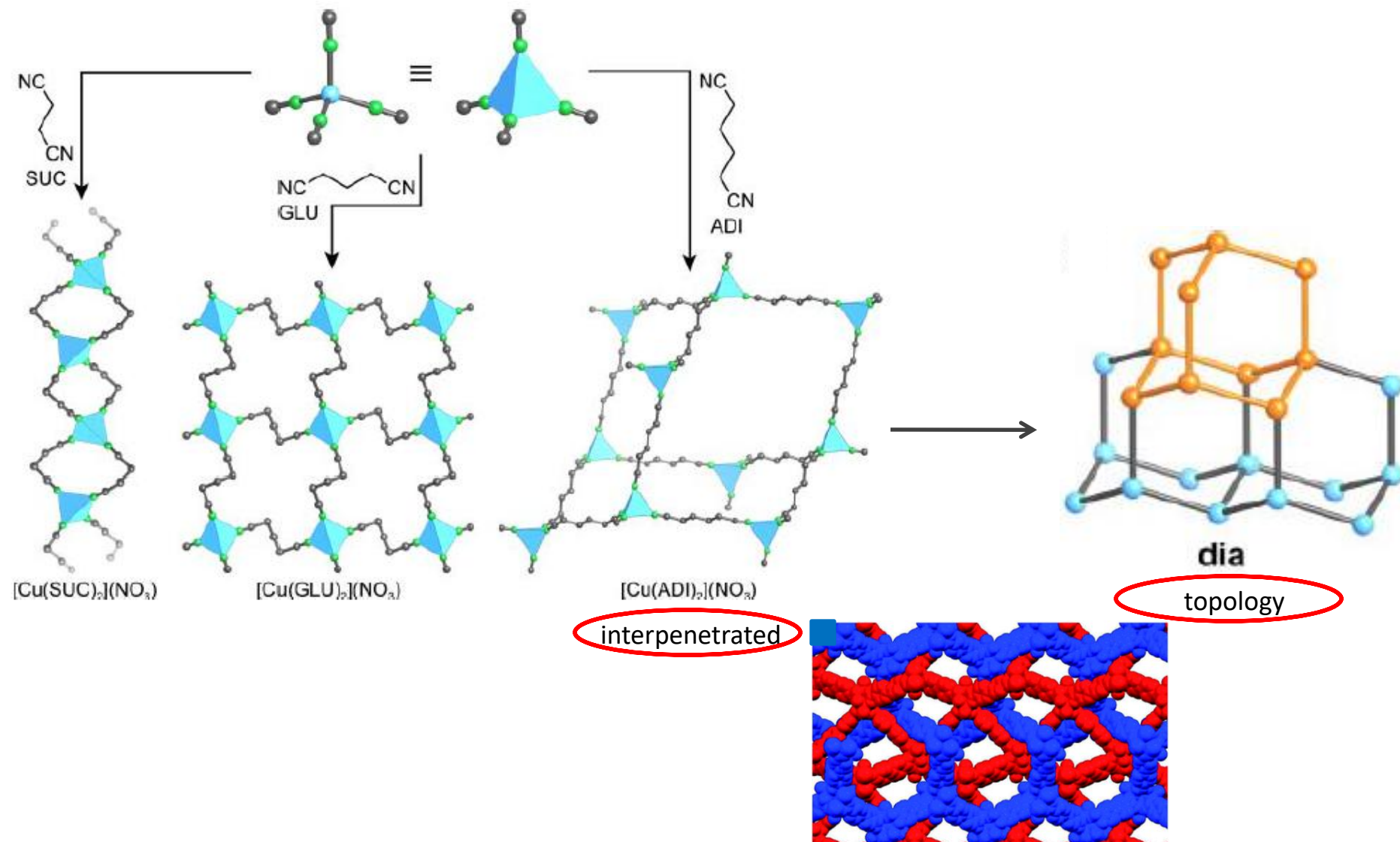


Extended structures
Molecular chemistry



Coordination Networks

(Introduction of nitriles for the link of ions)

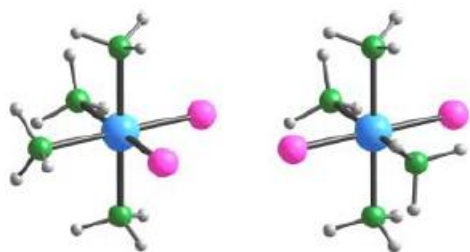


Y. Kinoshita, I. Matsubara and Y. Saito, *Bulletin of the Chemical Society of Japan*, **1959**, 32(7) 741–747.

Y. Kinoshita, I. Matsubara and Y. Saito, *Bulletin of the Chemical Society of Japan*, **1959**, 32(11) 1216–1221.

A. Wells, *Acta Crystallographica*, **1954**, 7 (8–9): 535–544.

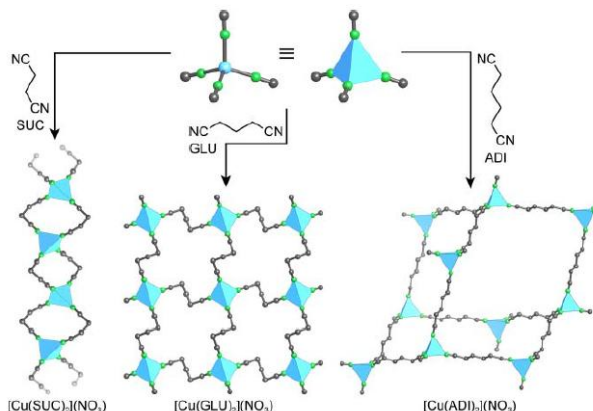
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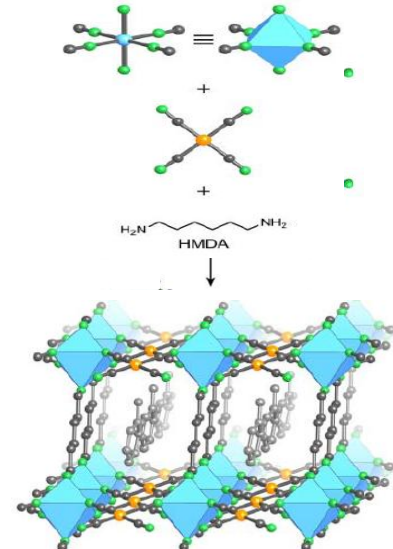
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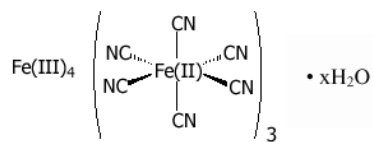
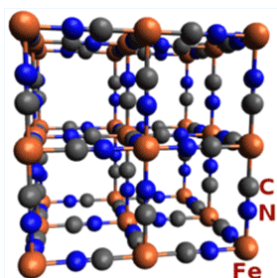
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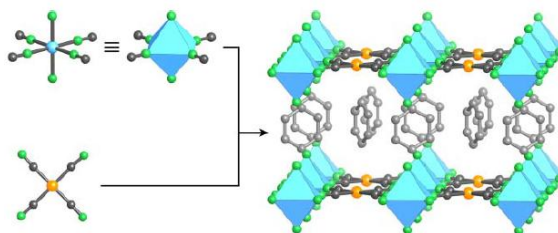
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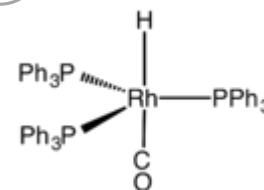
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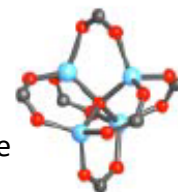


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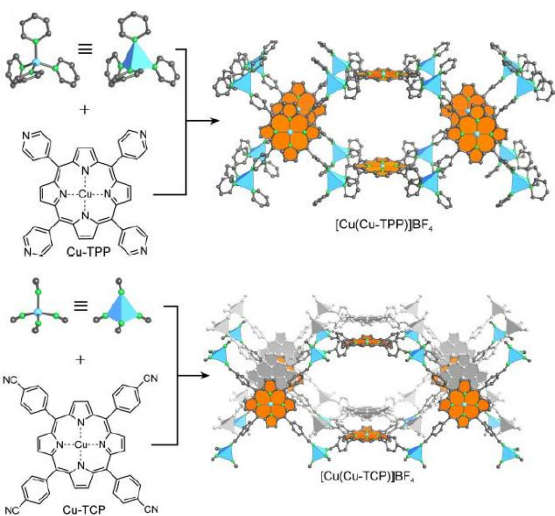
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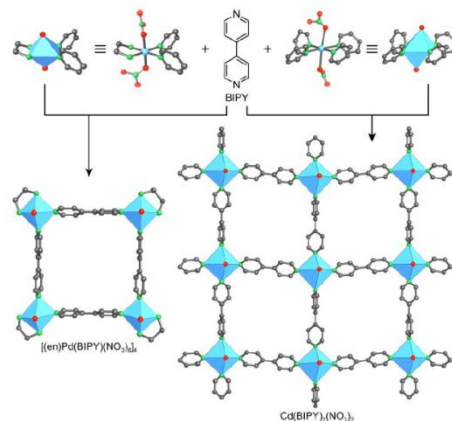
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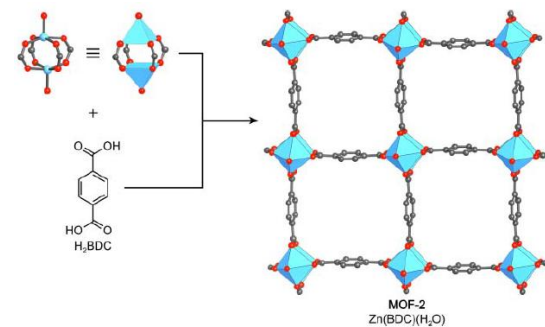


1990



Introduction of bipyridines for the link of ions (Fujitu)

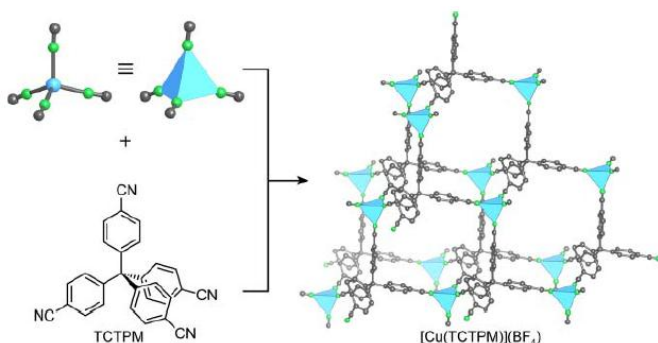
1998



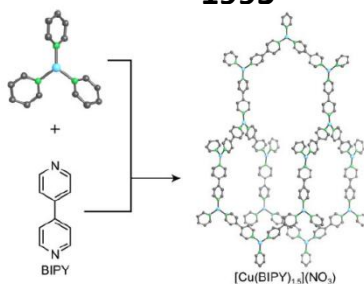
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Design of coordination networks (Robson & Hoskins)

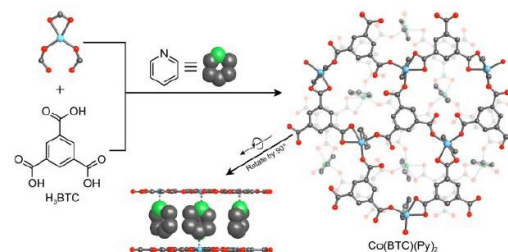
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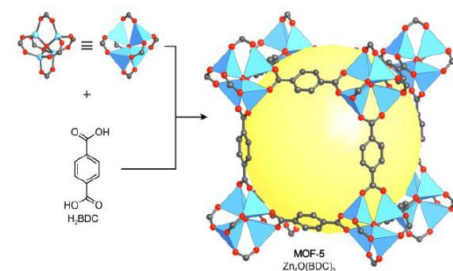


The term metal-organic framework is coined for the first time (Yaghi)



Introduction of Charged Linkers(Yaghi)

1999

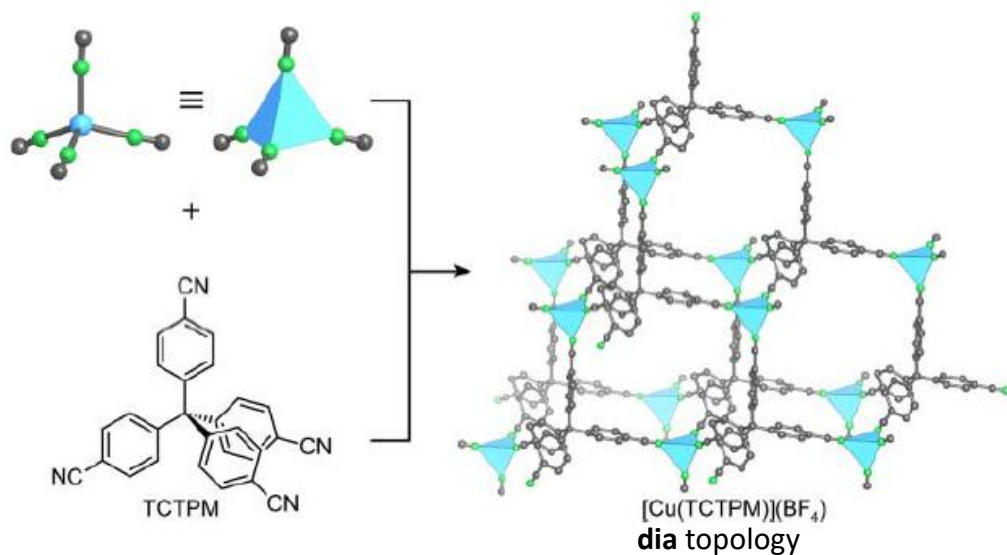
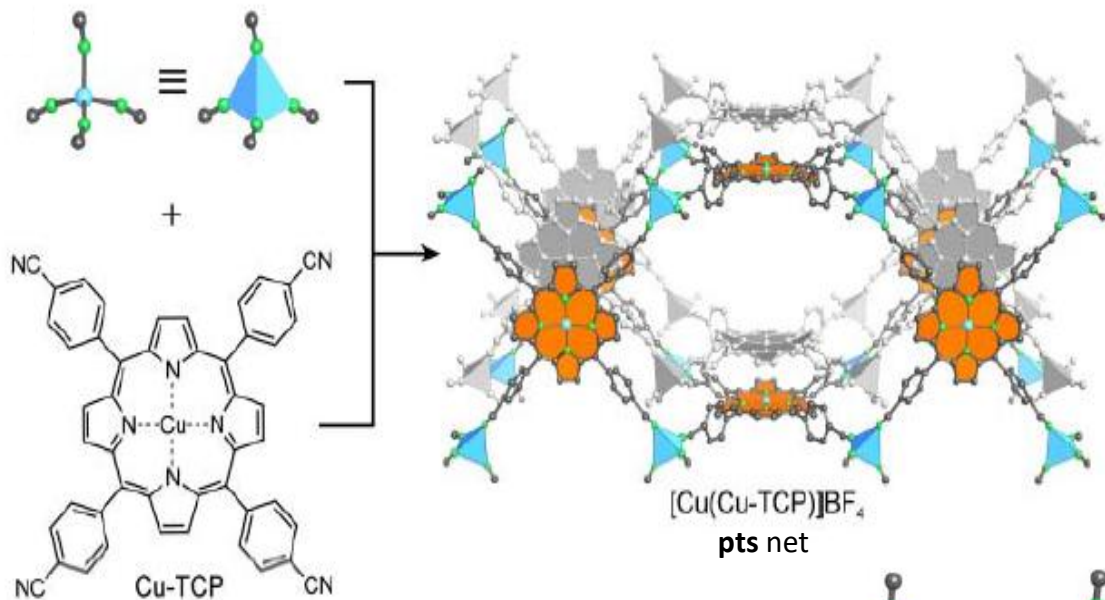


The first 3D MOF (Yaghi)



Coordination Networks

(Introduction of nitriles for the link of ions)



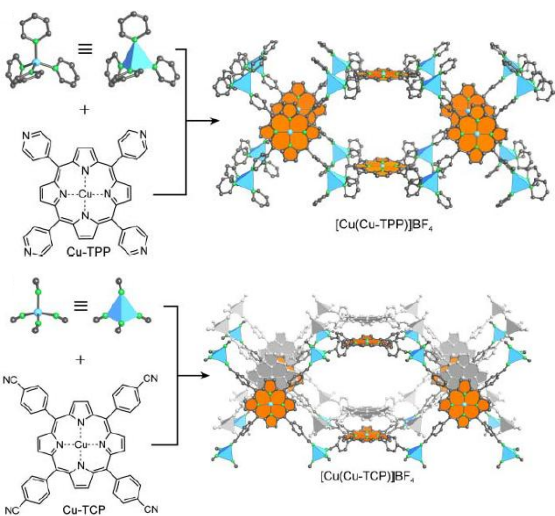
B.F. Hoskins and R. Robson, *Journal of the American Chemical Society*, **1989**, 111 5962–5964.

R.W. Gable, B.F. Hoskins and R. Robson, *Chemical Communications*, **1990**, 10 762–763.

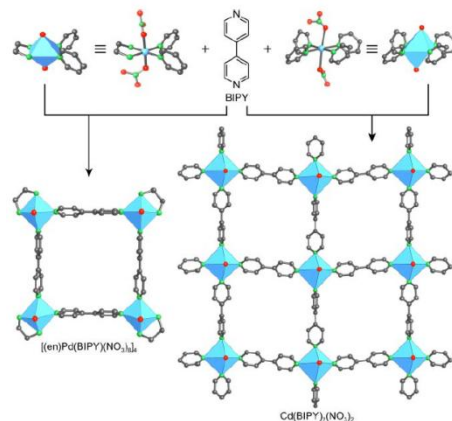
B.F. Abrahams, B.F. Hoskins, D.M. Michail and R. Robson, *Nature*, **1994**, 369 727–729



Emergence of MOFs

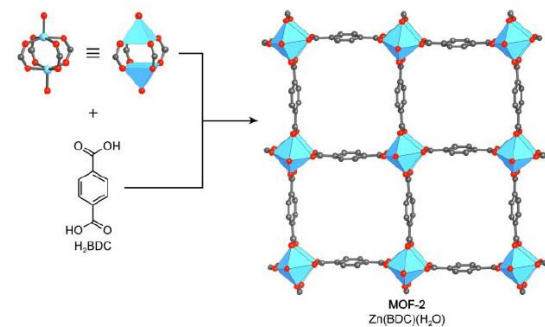


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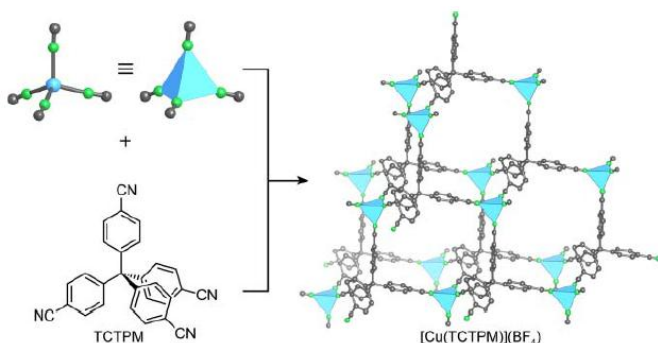
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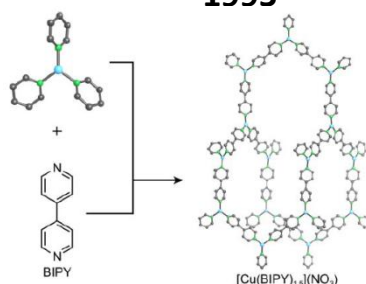
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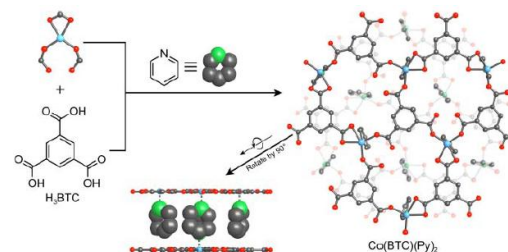
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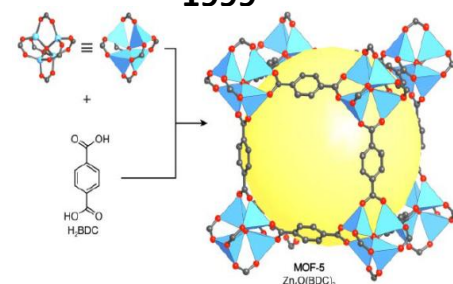


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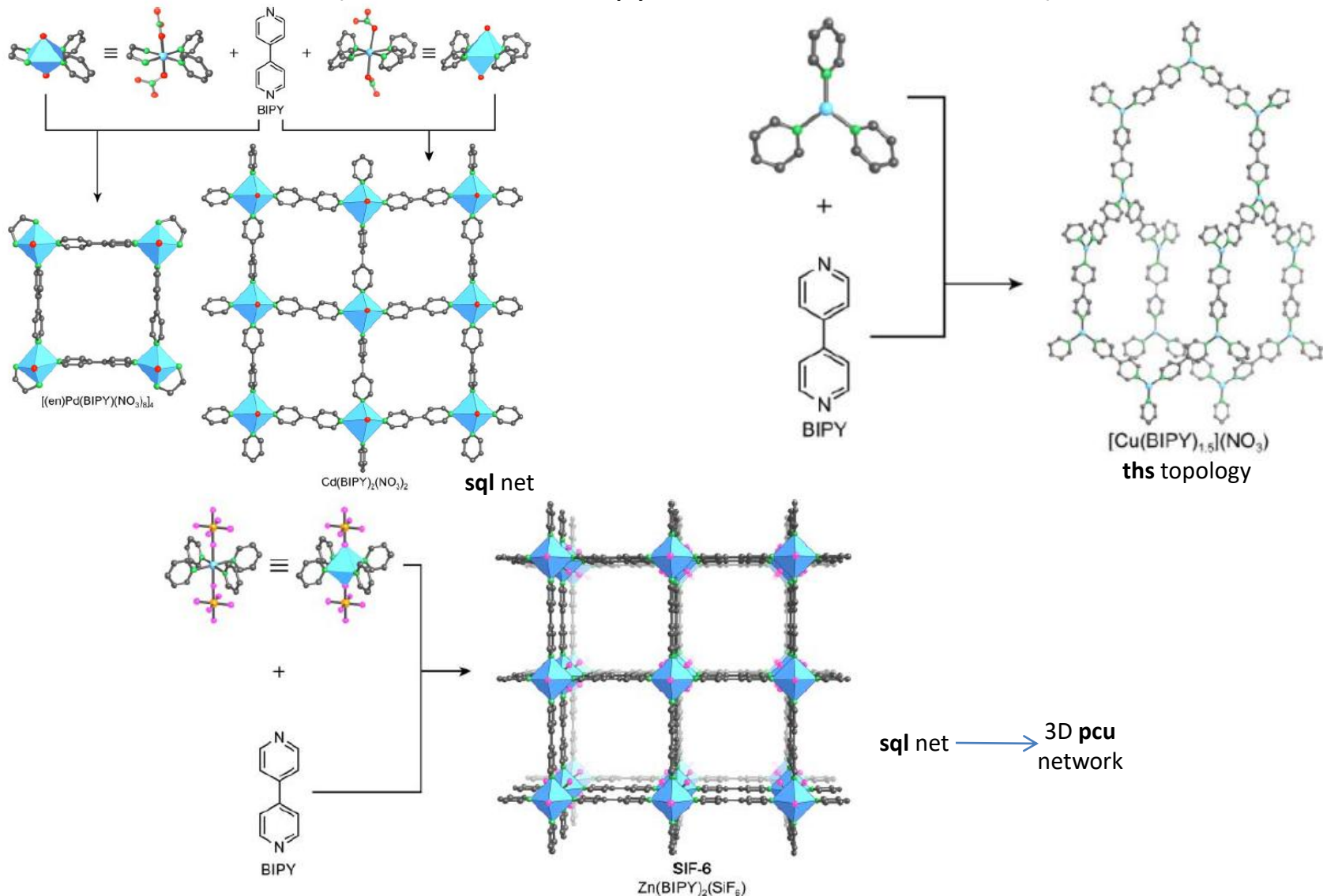


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Coordination Networks

(Introduction of bipyridines for the link of ions)



M. Fujita, J. Yazaki and K. Ogura, *Journal of the American Chemical Society*, **1990**, 112 5645–5647.

O. Yaghi and H. Li, *Journal of the American Chemical Society*, **1995**, 117 10401–10402.

S. Subramanian and M.J. Zaworotko, **1995**, *Angewandte Chemie*, 34 2127–2129.



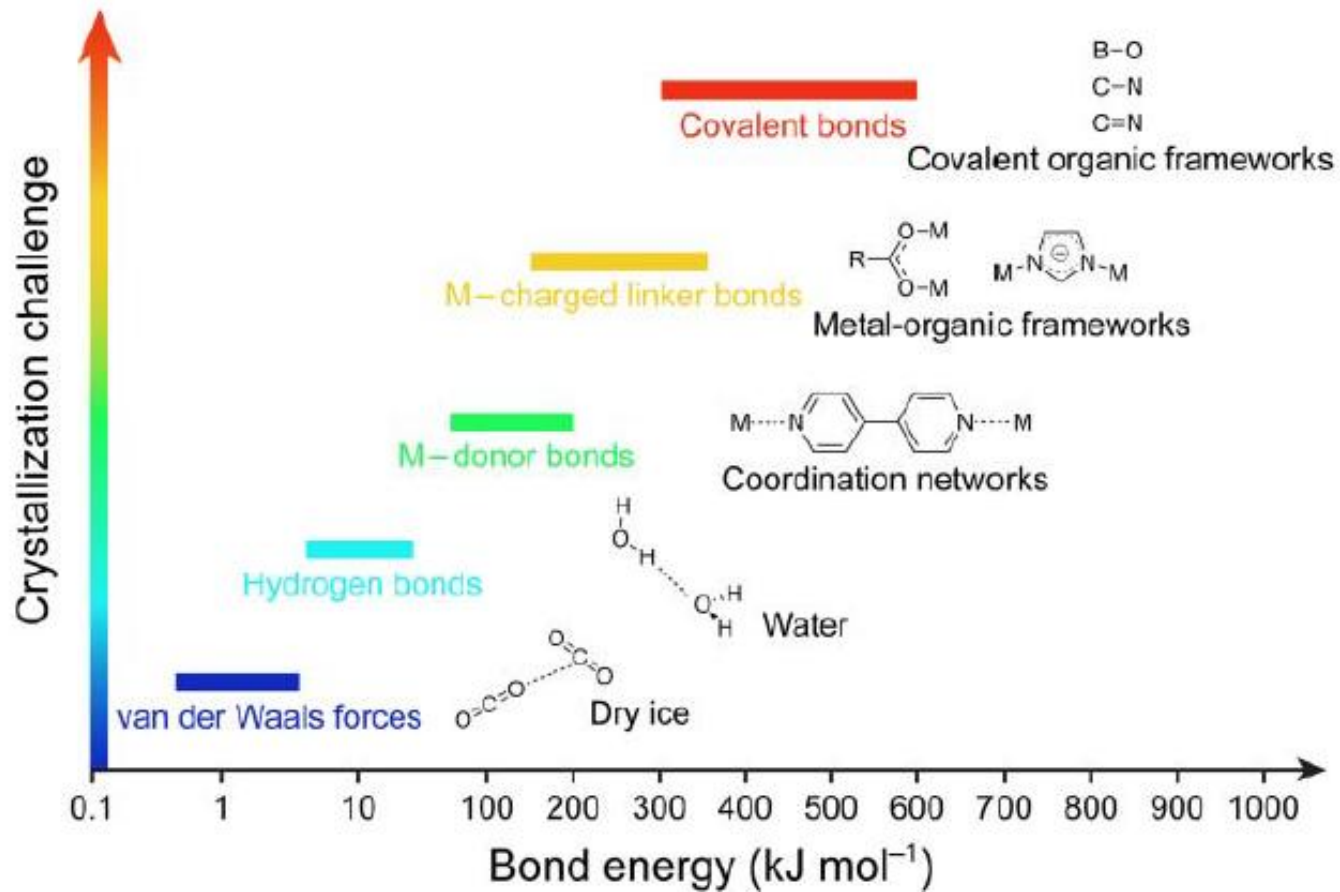
Coordination Networks

The aforementioned design principles can be used to construct a wide variety of coordination networks through the judicious choice of metal ions and organic linkers, **BUT**

- Collapsed upon removal of solvents or exchange of ions in pores
- Not Designable because they are made of single metal ion vertex
- Chemical instability, because of weak M-N(pyridine, nitrile) –type of links

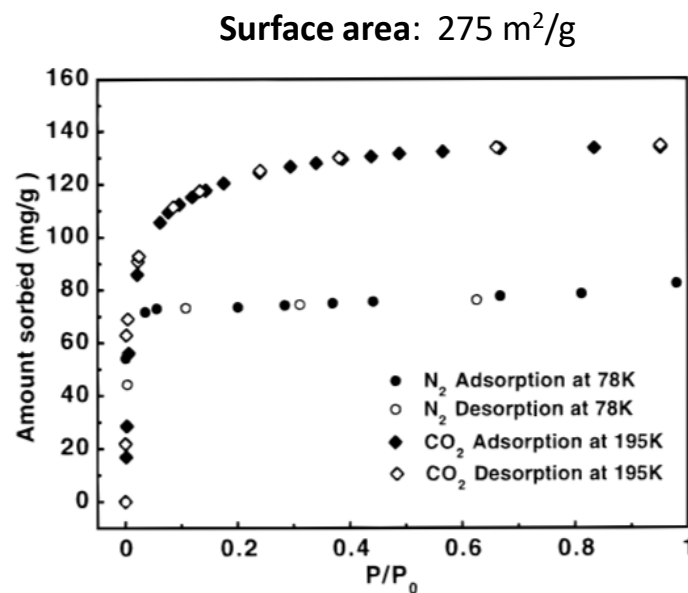
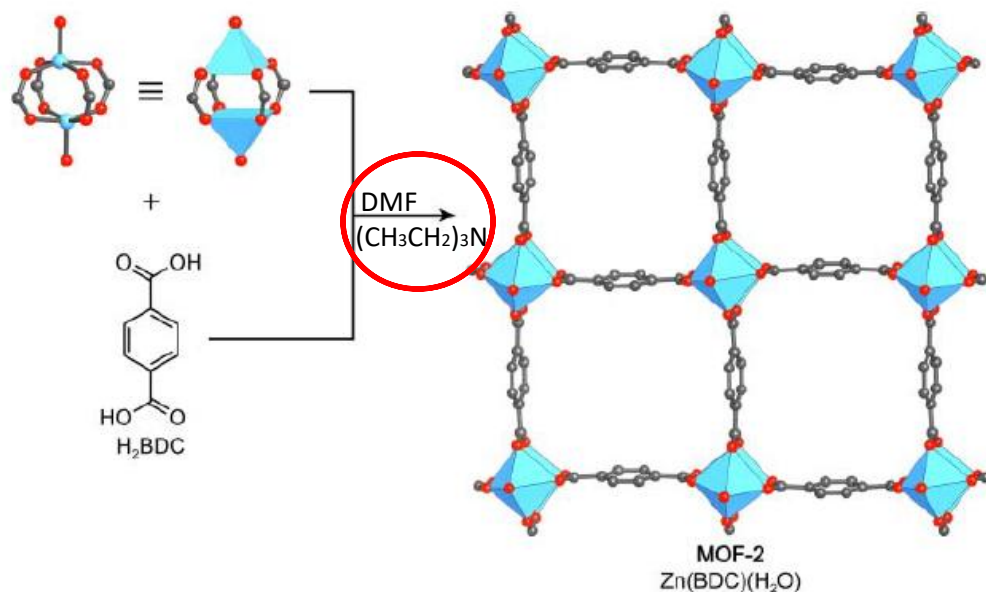


Crystallization problem



MOF-2

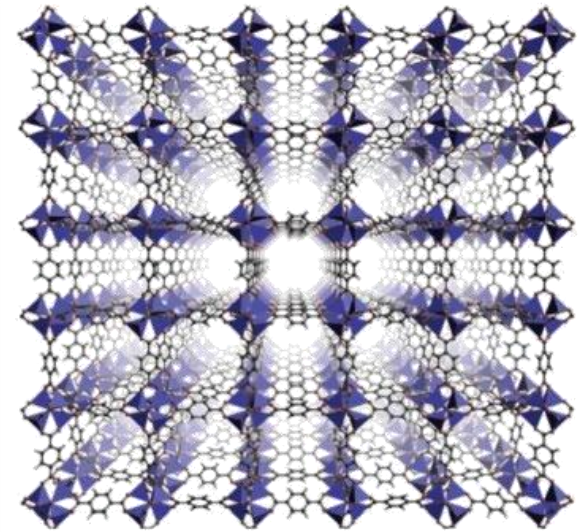
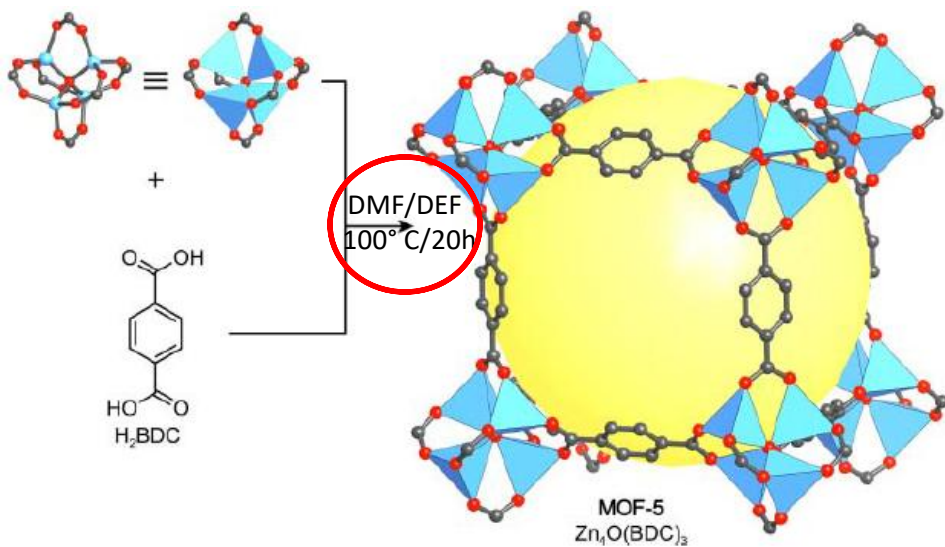
(Introduction of SBUs and charged linkers) for the link of ions



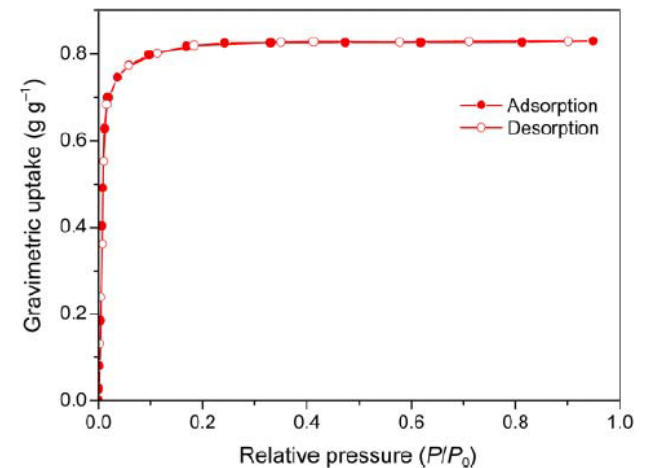
H. Li, M. Eddaoudi, T.L. Groy and O. M. Yaghi, *Journal of the American Chemistry Society*, **1998**, 120 (33): 8571–8572.



MOF-5

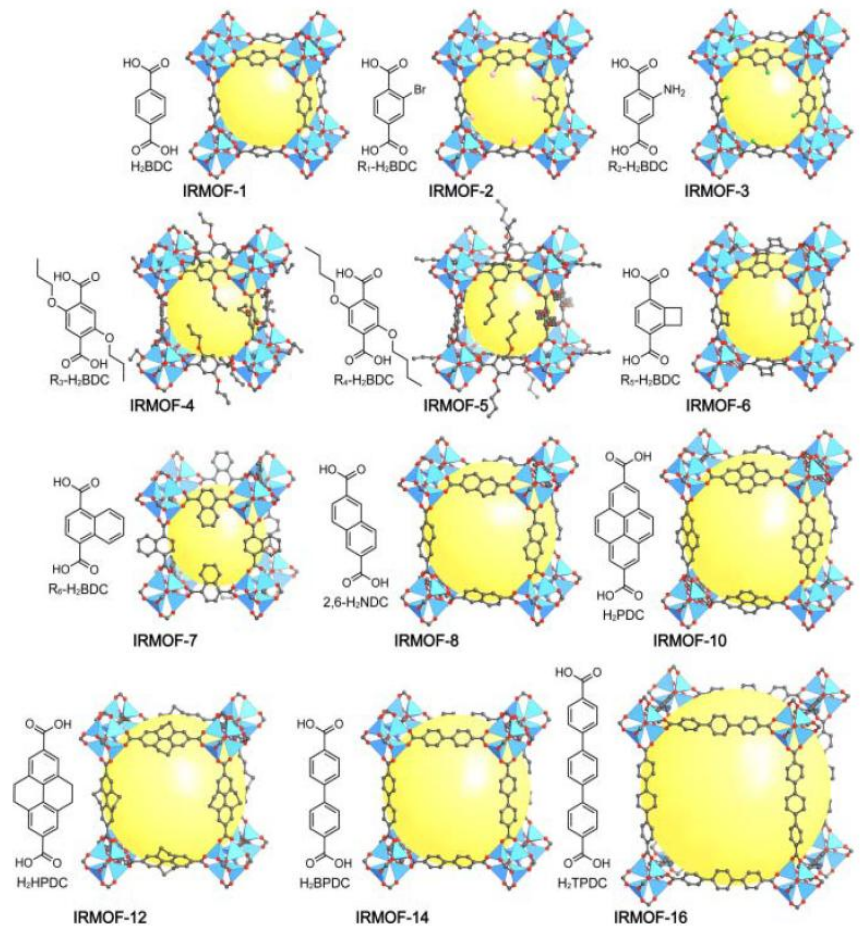
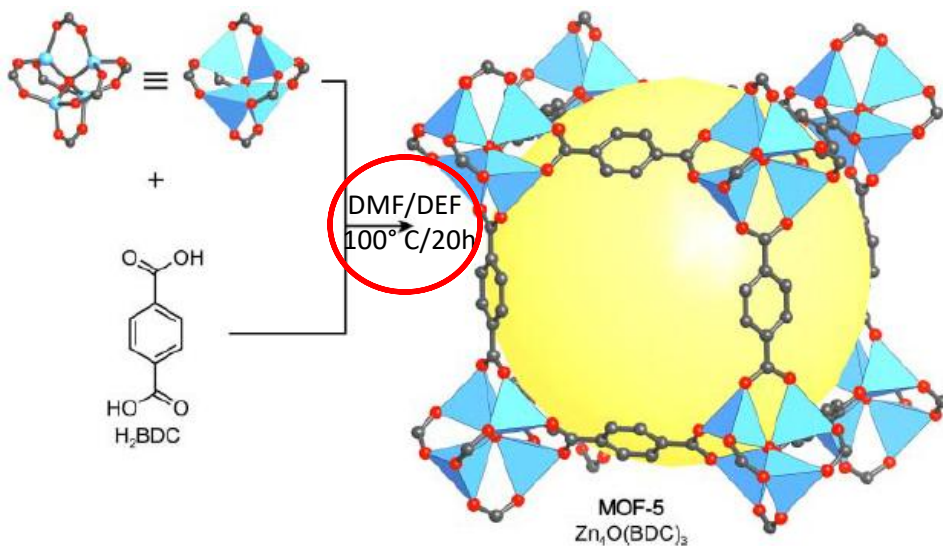


Surface area: 2900 m²/g



MOF-5

Isorecticular principle: elongation(pore opening) or functionalization of the linker without modifying its general shape and connectivity



M. Eddaoudi, J. Kim, N. Rosi, D. Vodak, J. Wachter, M. O'Keeffe and O. M. Yaghi, *Science*, **2002**, 295 (5554): 469–472.



- “One of the continuing scandals in the physical sciences is that it remains general **impossible to predict the structure** of even the simplest crystalline solids from a knowledge of their chemical composition”

John Maddox, *Nature*, **1988**, 335, 201.

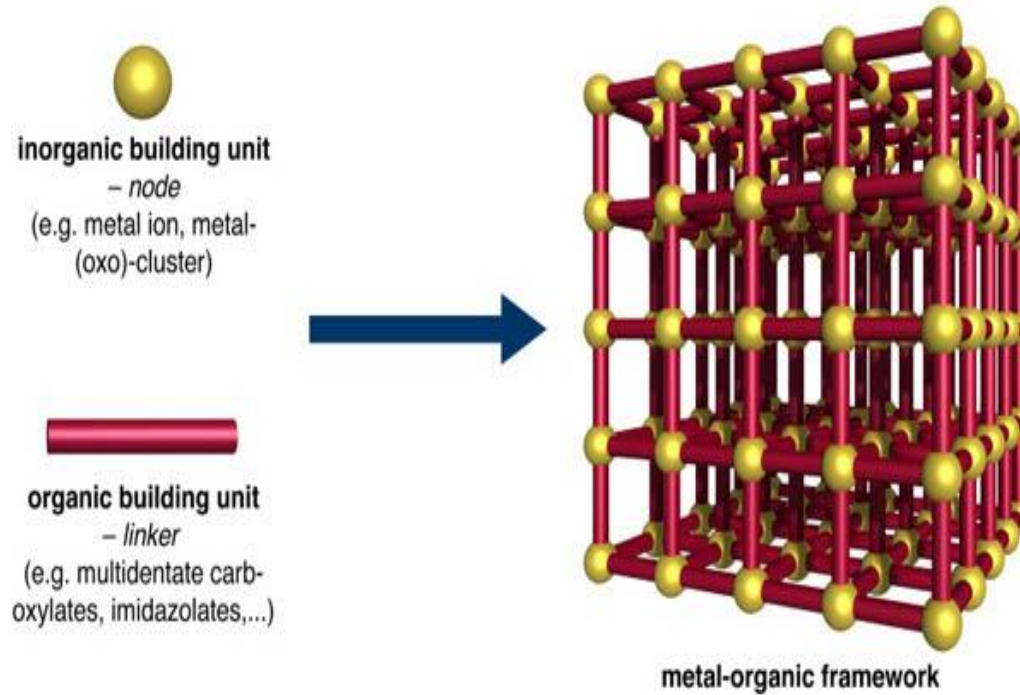
- “Organic chemists are masterful at exercising control in zero dimension. One subculture of organic chemists has learned to exercise control in one dimension. These are polymer chemists, the chain builders...But **in two or three dimensions, it is a synthetic wasteland.**”

R.Hoffmann, *Scientific American*, **1993**, 65-73.

Reticular chemistry: is the study of linking **molecular building blocks** by strong to make **crystalline** extended structures.



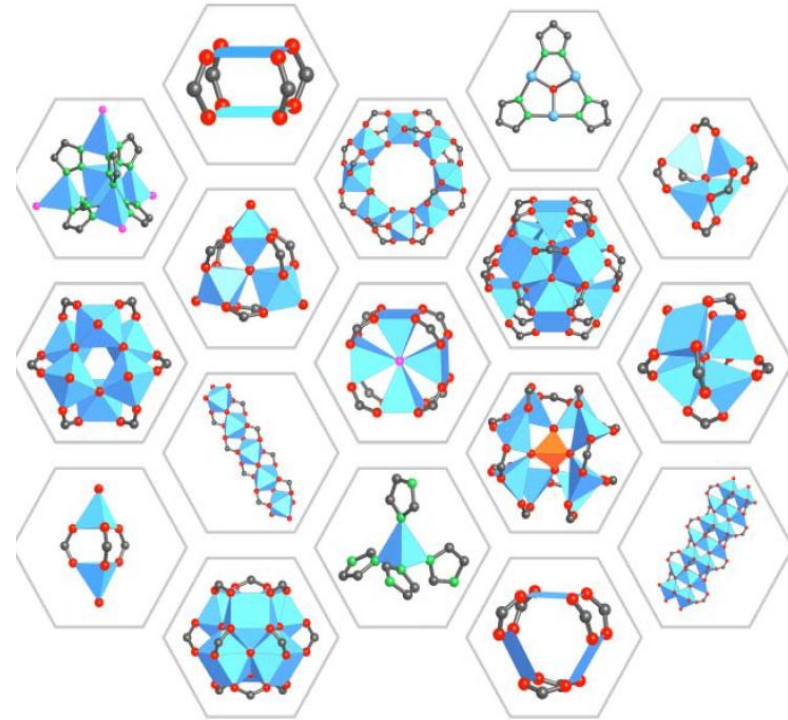
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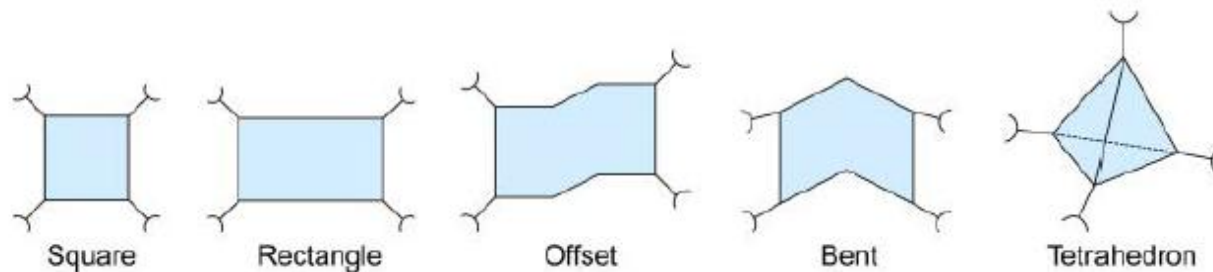
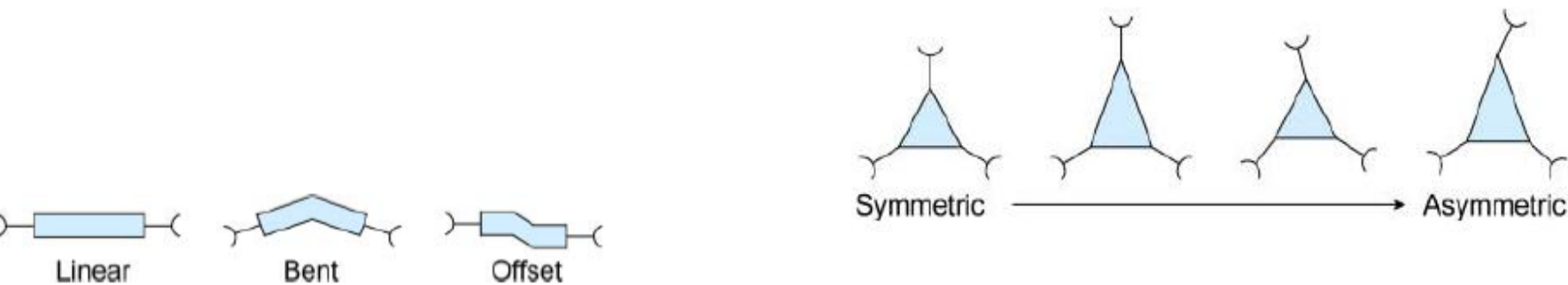
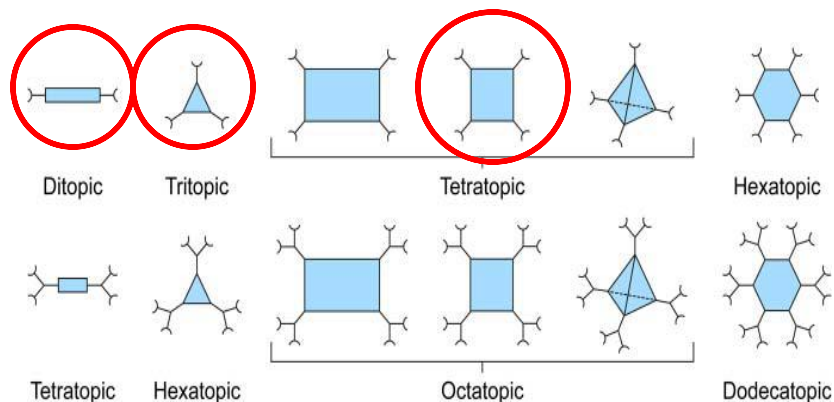
Design

(Secondary Building Units, SBUs)

H																	He
Li	Be							B	C	N	O	F	Ne				
Na	Mg							Al	Si	P	S	Cl	Ar				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	*	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	**	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og
			La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
			Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	Ho

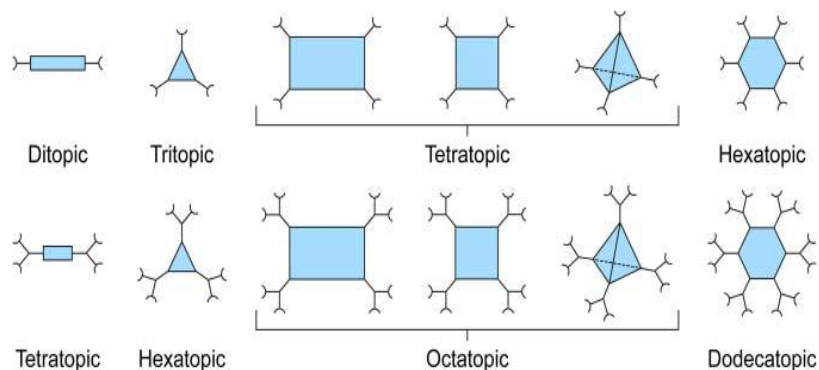


Design (Organic linker)

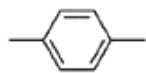


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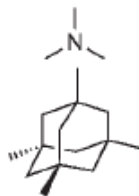
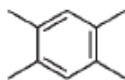
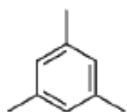
(Organic linker)



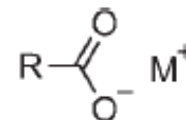
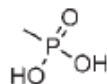
Extending units



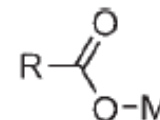
Core units



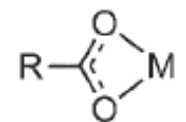
Binding groups



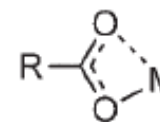
Ionic



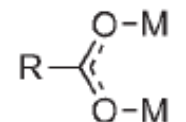
Unidentate



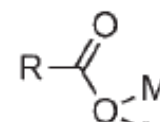
Chelating (sym.)



Chelating (asym.)

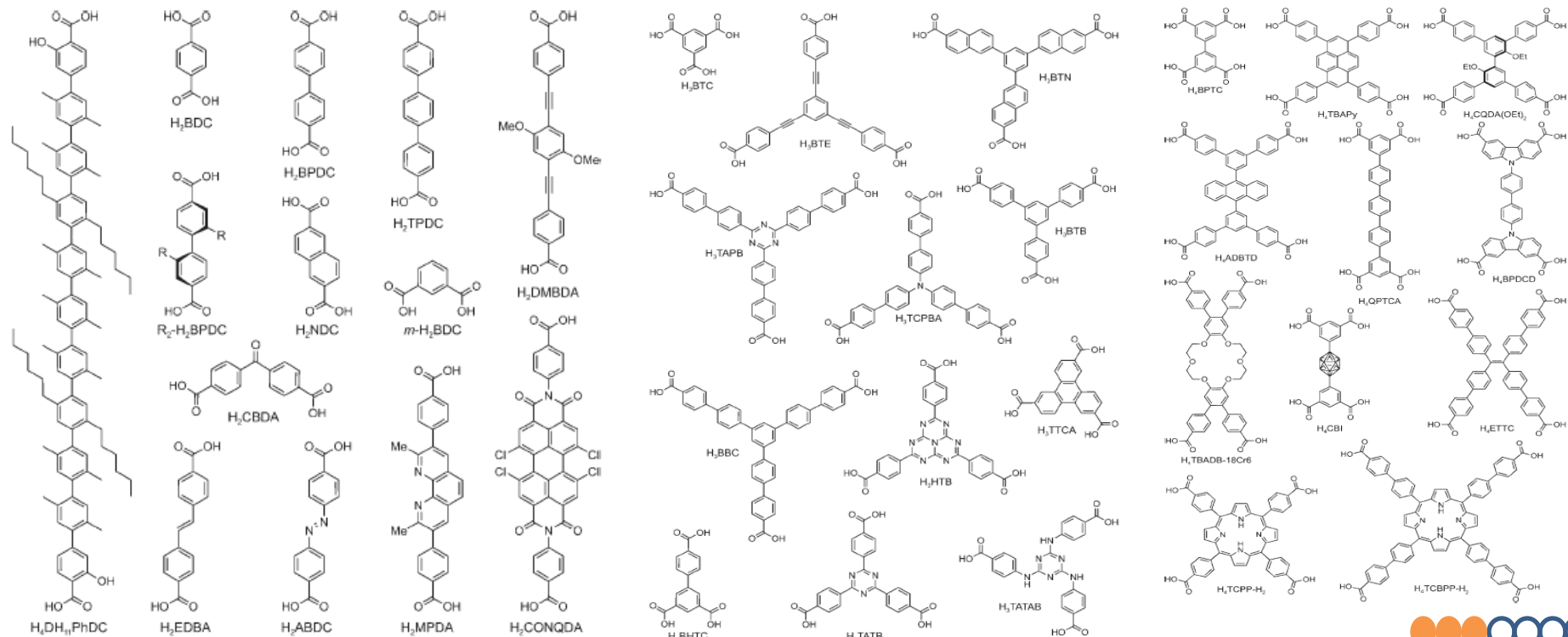
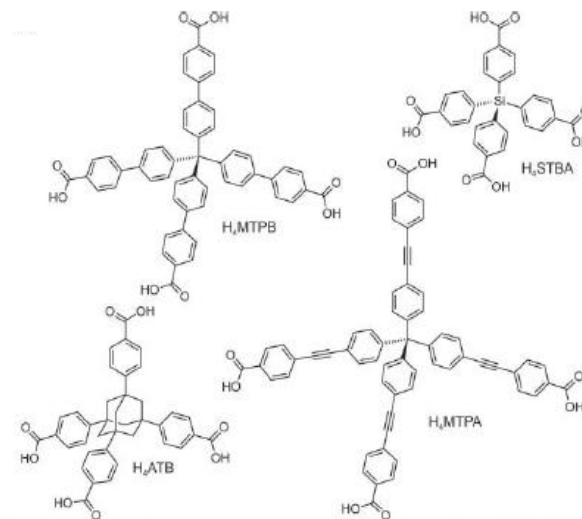
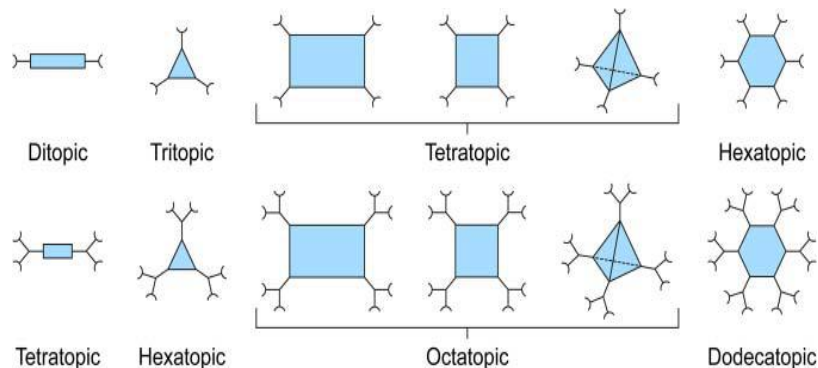
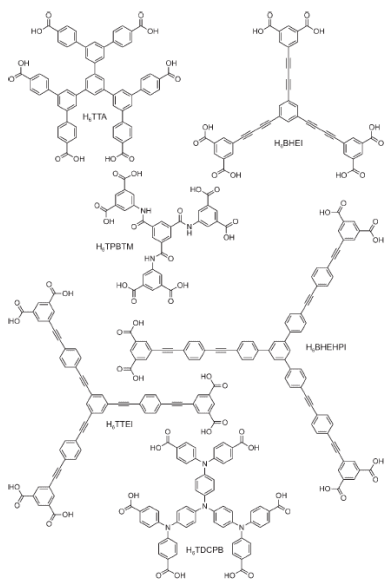


Bridging (syn-syn)



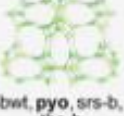
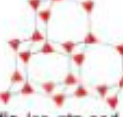
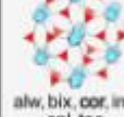
Bridging (monoatomic)

Design (Organic linker)



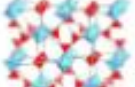
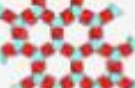
Design

(Application of SBU approach)

Building unit 2 \ Building unit 1	2-c Linear	3-c Triangle	4-c Square	4-c tet	6-c Hexagon	6-c oct
3-c Triangle	 srs	 bwt, pyo, srs-b, ths-b	 fjh, fmj, gee, iab, yac, yao	 asn, ept, ofp	 cys, dnf*	 anh, ant, apo, brk, cep*, cml, czz, eea, qom, rti, tsx, zzz
4-c Square	 nbo, lvt, rhr	 pto, tbo	 cev, cdl, cdm, cdn, cds, cdz, mot, muo, qdl, qzd, ssd, sse, ssf, sst	 pts	 nts	 myd, ybh
4-c tet	 dia, lcs, qtz, sod	 bor, ctn	 fgl, mog, pds, pth, pti, ptr, ptt	 bni, byl, cag, cbt, coe, crb, fel, icm, kea, lon, pci, qtz-b, sca, tpd, ucn	—	 alw, bix, cor, ing, spl, toc
6-c Hexagon	 hxg	 cys, dnf	 she	—	 hxg-b	—

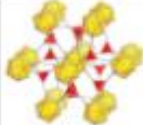
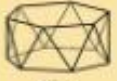
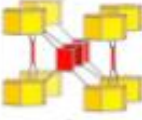
Design

(Application of SBU approach)

Building unit 1 \ Building unit 2	2-c Linear	3-c Triangle	4-c Square	4-c tet	6-c Hexagon	6-c oct
6-c oct	 pcu, bcs, crs, reo	 pyr, spn	 soc	 gar, iac, ibd, toc	-	 pcu-b, bcs-b
6-c trp	 lcy, acs	 ceq, dag, fmz, hwx, moo, sab, slit, ydq	 stp	 fsi, hea, tpt	 htp	 nia
8-c cub	 bcu	 the	 scu, csq, sqc	 flu	-	 ocu
12-c cuo	 fcu	 sky	 ftw	 edc	-	-

Design

(Application of SBU approach)

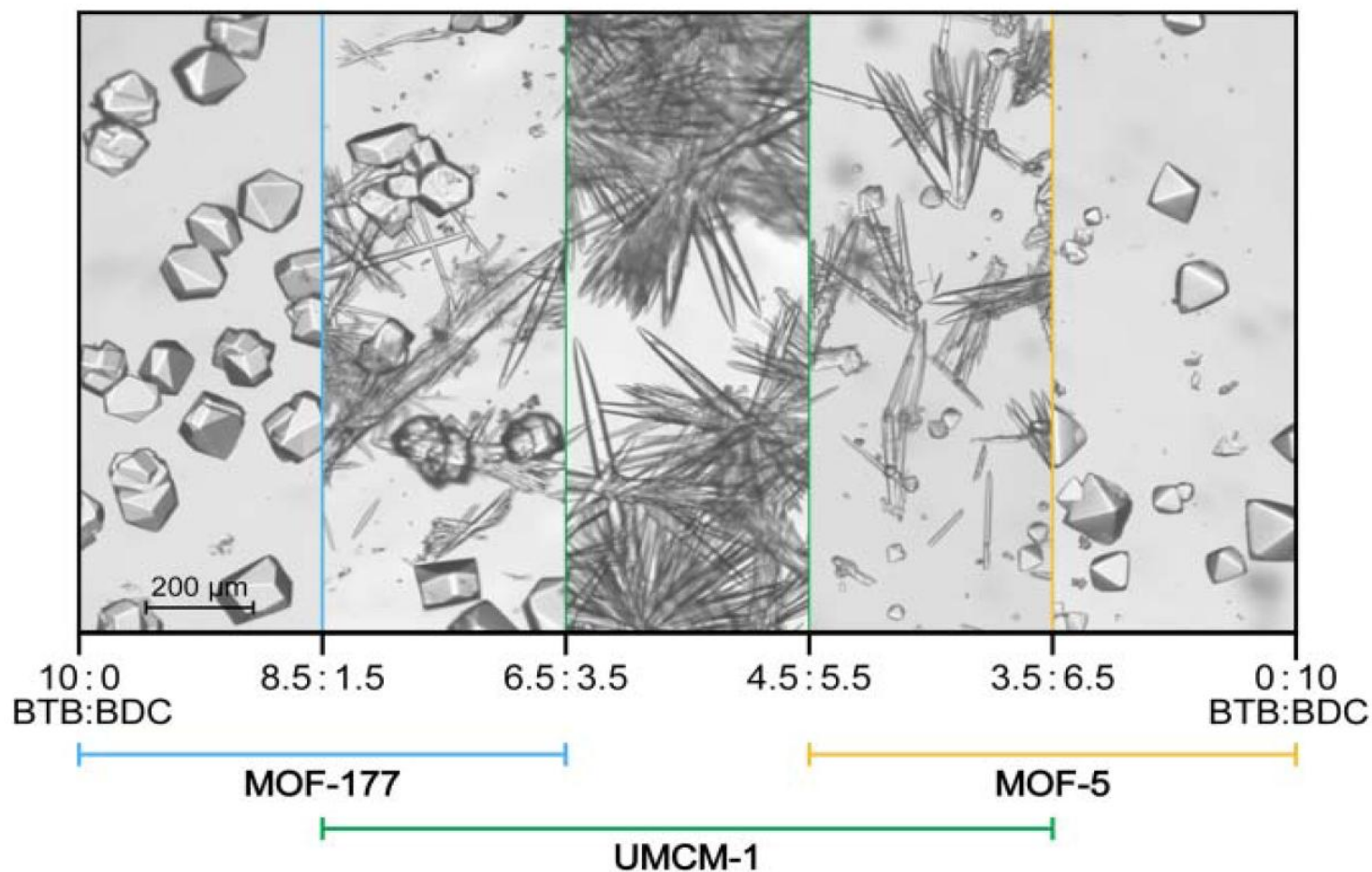
Building unit 1 \ Building unit 2	 2-c Linear	 3-c Triangle	 4-c Square	 4-c tet	 6-c Hexagon	 6-c oct
 12-c ico	-	-	-	 lth	-	-
 12-c hpr	-	 aea	 shp	-	-	-
 12-c tte	-	 ttr	-	-	 mgc	-
 24-c tro	-	-	-	 twf	-	-

Design

(Complexity and Heterogeneity)

Complexity

- (i) multiple different linkers
- (ii) multiple different SBUs
- (iii) multiple different SBUs and linkers

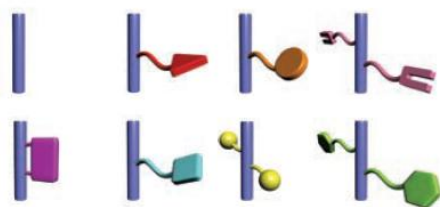


Design

(Complexity and Heterogeneity)

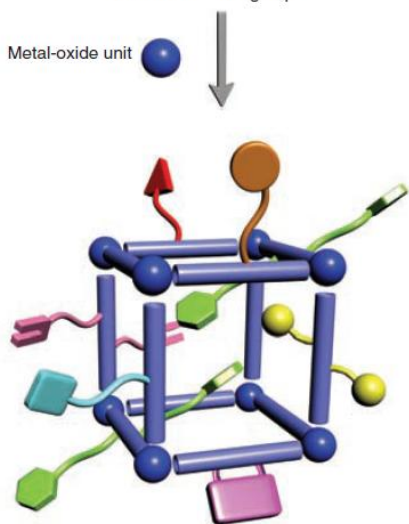
Complexity

- (i) multiple different linkers
- (ii) multiple different SBUs
- (iii) multiple different SBUs and linkers



Phenylene units with various covalently bound functional groups

Metal-oxide unit

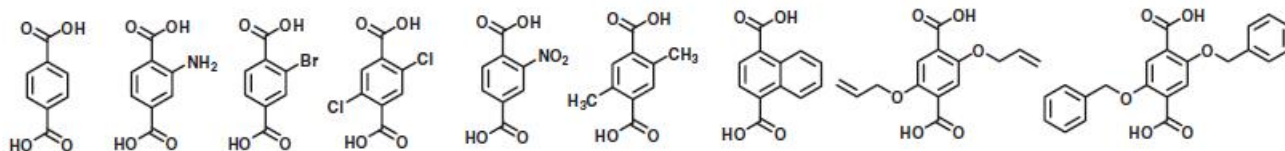
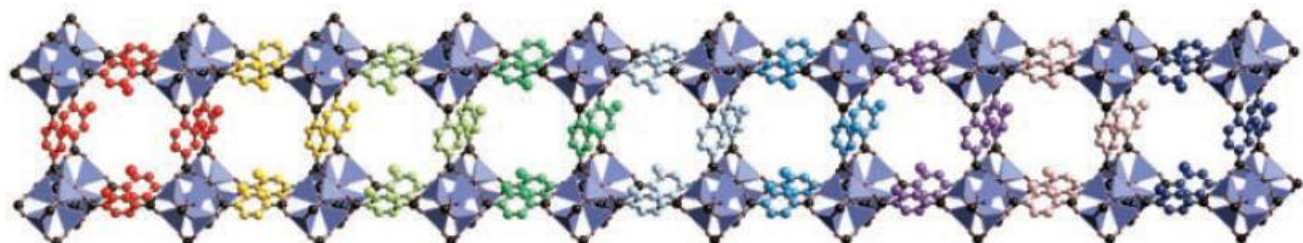


MTV-MOF-5 structure
with eight different functionalities

Heterogeneity

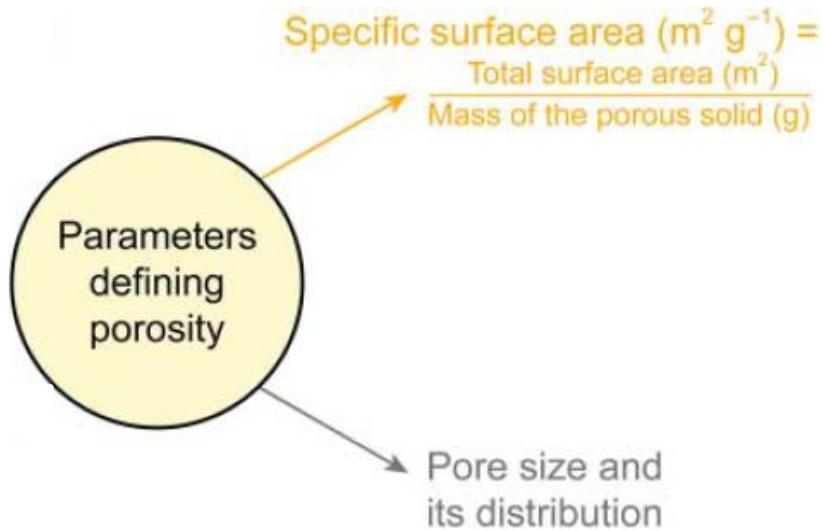
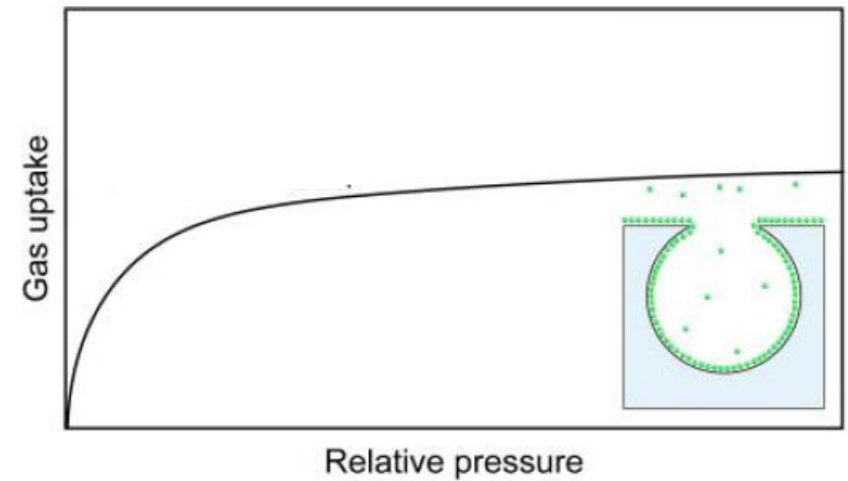
- (i) multiple interchangeable linkers that are identical in terms of their binding groups and metrics but different with respect to their chemical composition
- (ii) multiple metal ions that form the same SBU

MTV-MOF-5



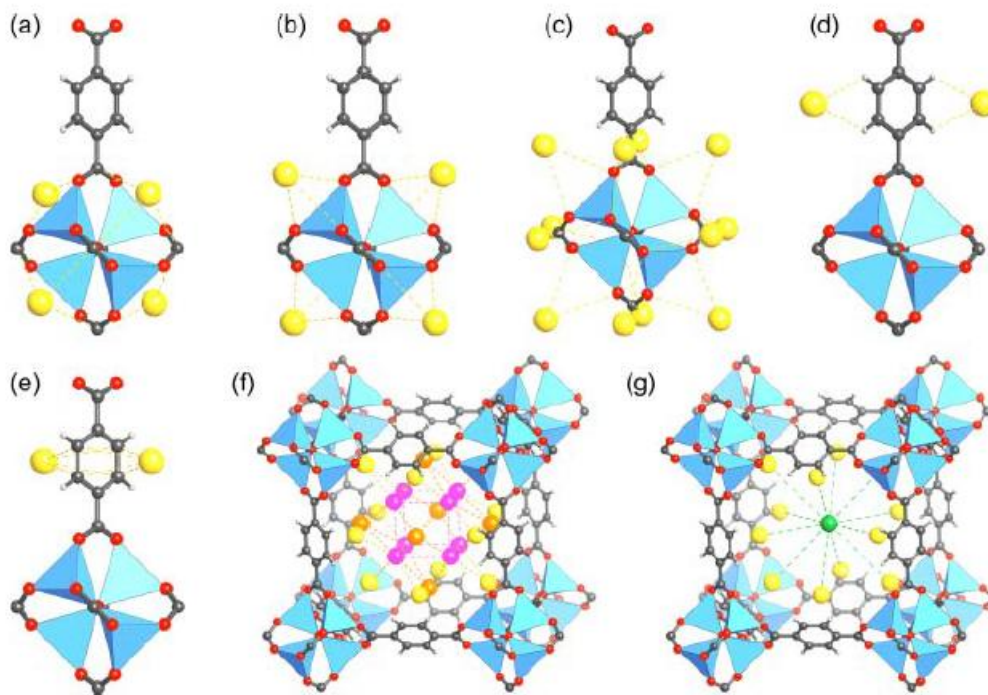
Design

(Fine-tuning of the sorption properties)



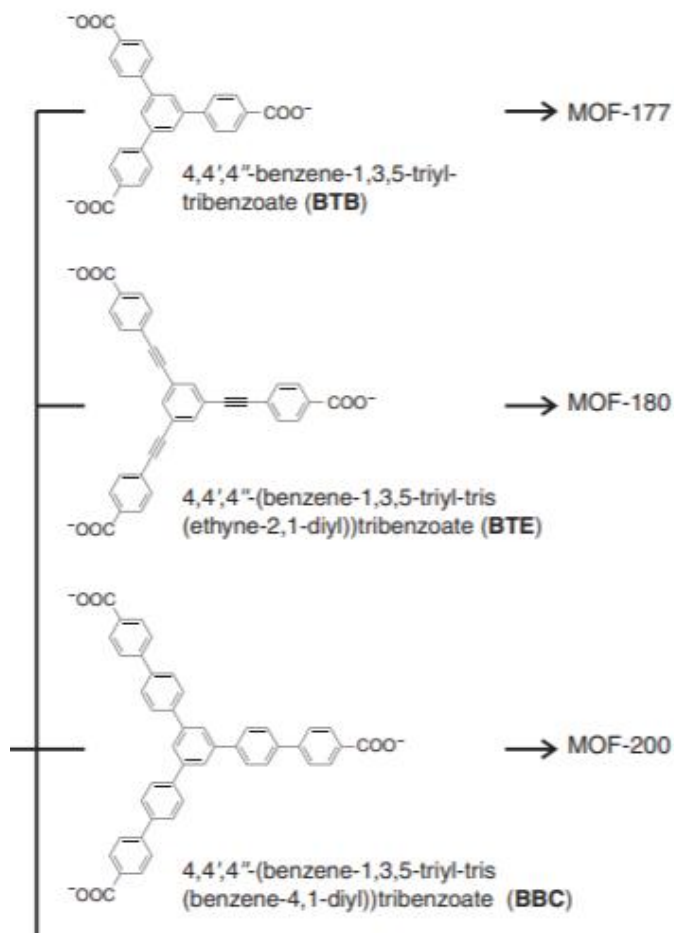
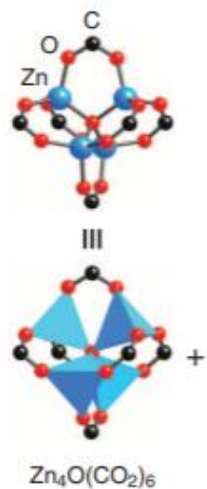
Design

(Ultra-high Surface Area)



Design

(Ultra-high Surface Area)

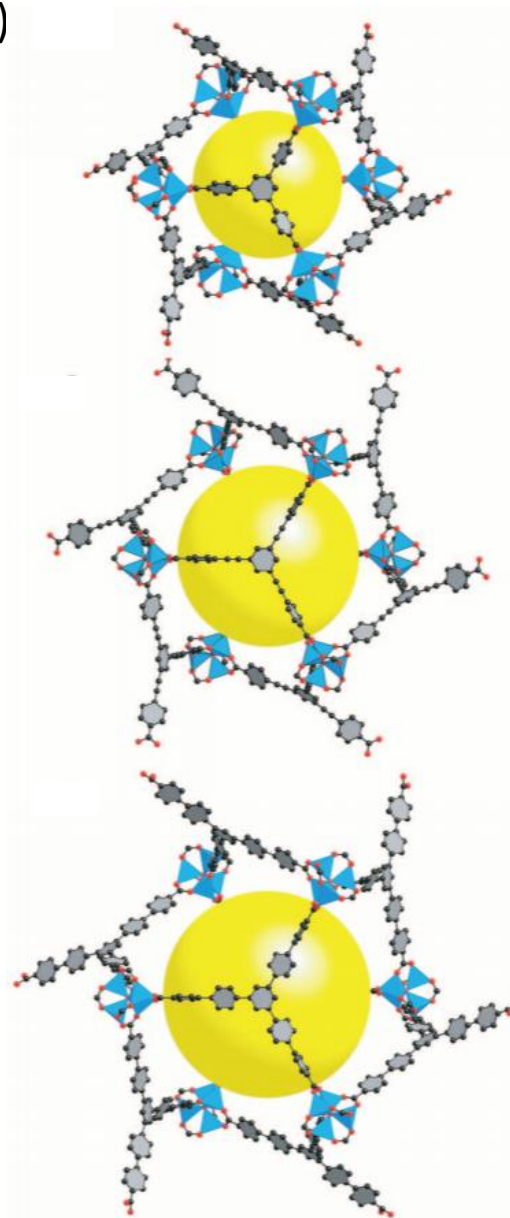


BET(m^2/g)

4530

4460

6240



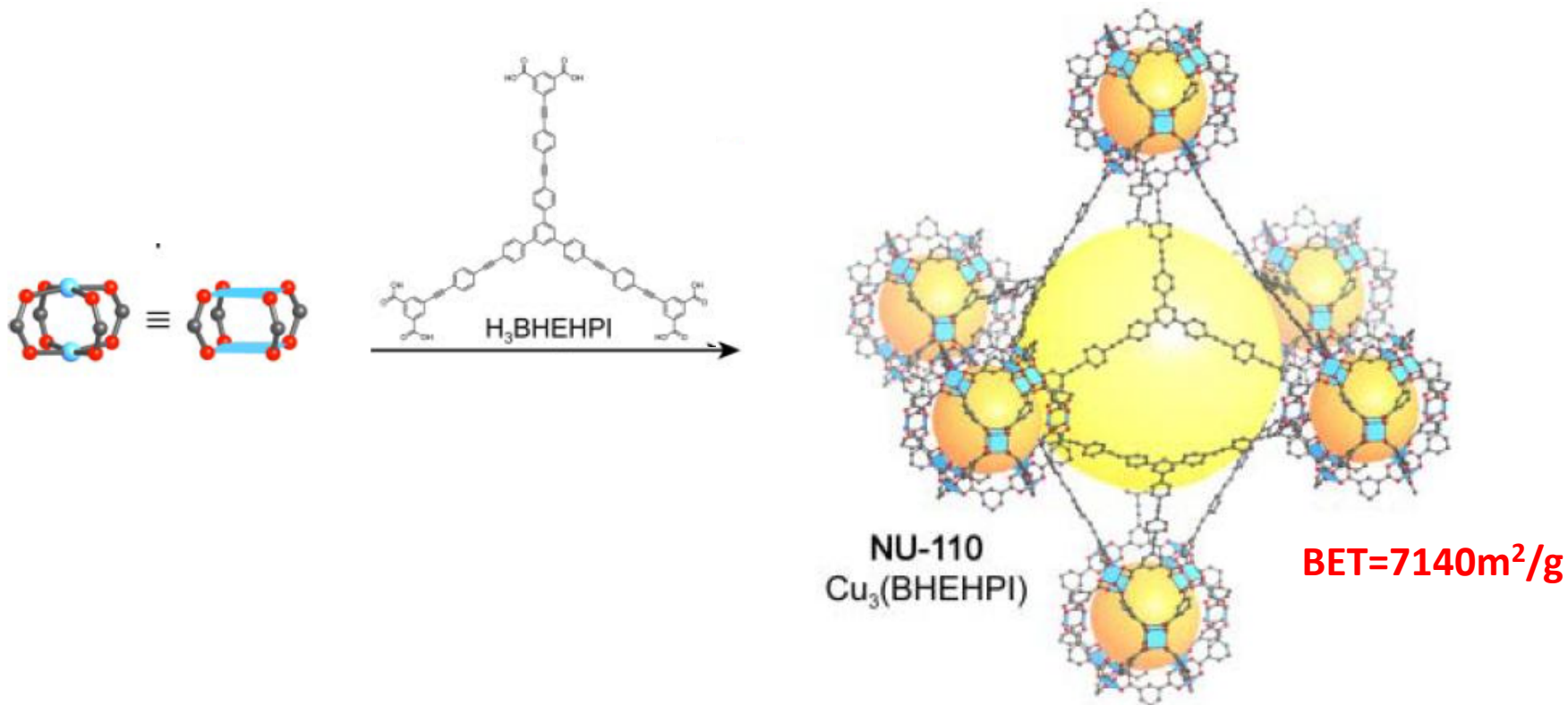
H. Furukawa, N. Ko, Y. B. Go, N. Aratani, S. Beom Choi, E. Choi, A. Ö. Yazaydin, R.Q. Snurr, M. O'Keeffe, J. Kim, O. M. Yaghi, *Science*, **2010**, 329(5990), 424–428.



Design

(Ultra-high Surface Area)

Metal–Organic Framework Materials with Ultrahigh Surface Areas. **Is the Sky the Limit?**

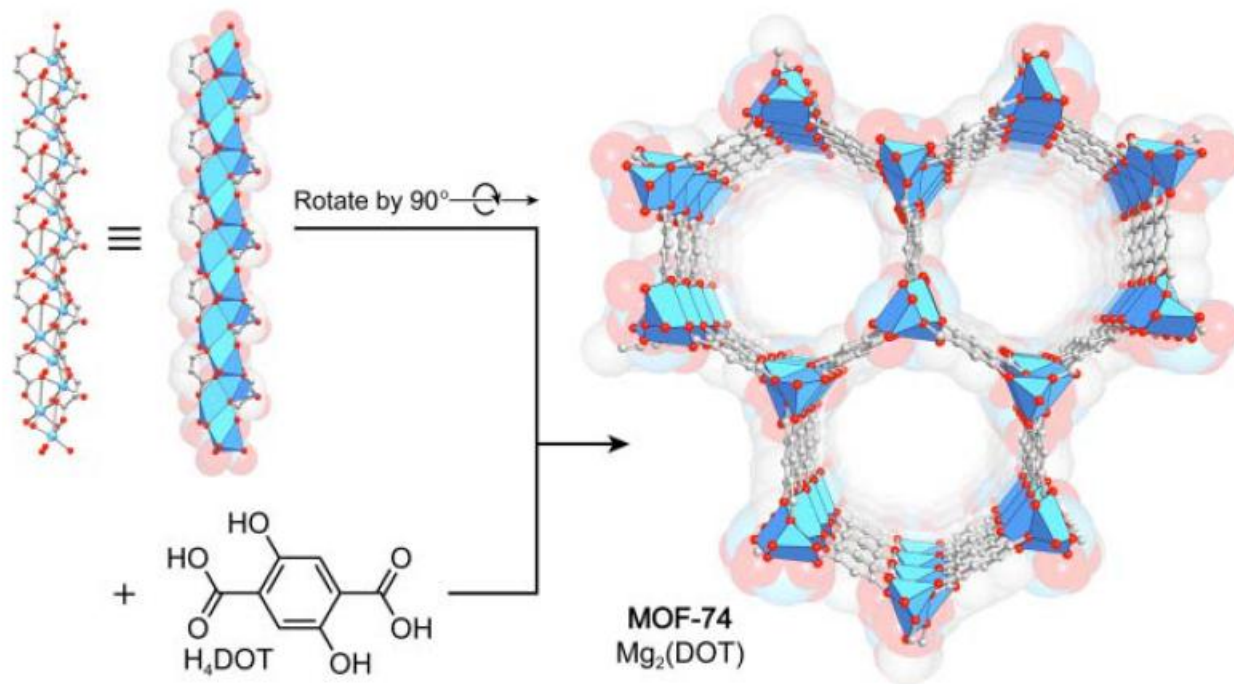


O. K. Farha, I. Eryazici, N. Cheon Jeong, B. G. Hauser, C. E. Wilmer, A. A. Sarjeant, R.Q. Snurr, S. T. Nguyen, A. Ö. Yazaydın and J. T. Hupp, *JACS*, **2012**, 134, 36, 15016-15021.



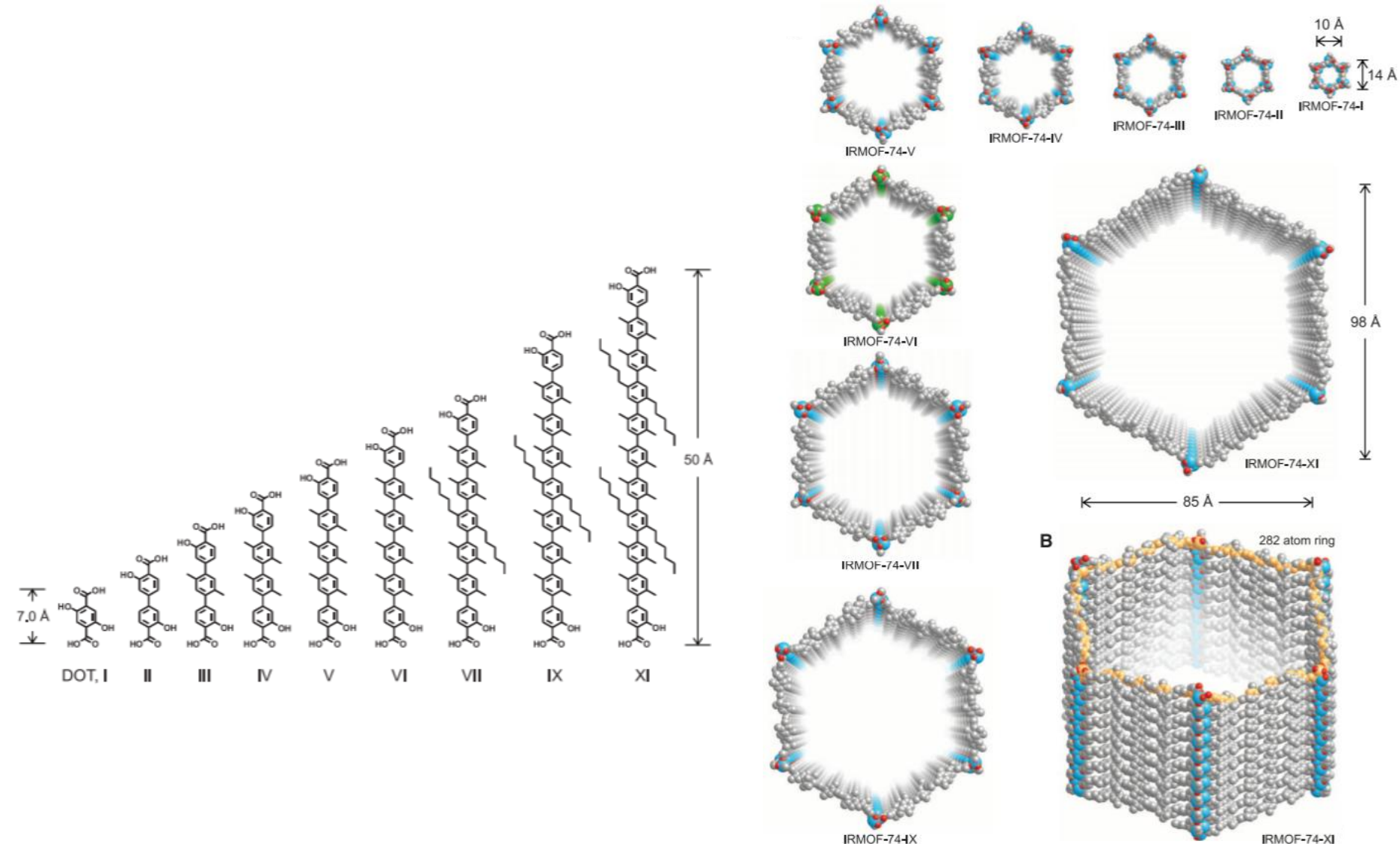
Design

(Pore opening)



Design

(Pore opening)

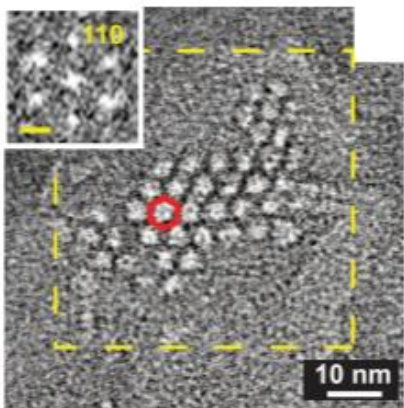


H. Deng, S. Grunder, K. E. Cordova, C. Valente, H. Furukawa, M. Hmadeh, F. Gándara, A. C. Whalley, Z. Liu, S. Asahina, H. Kazumori, M. O’Keeffe, O. Terasaki, J. F. Stoddart and O. M. Yaghi, *Science*, **2012**, 336(6084).

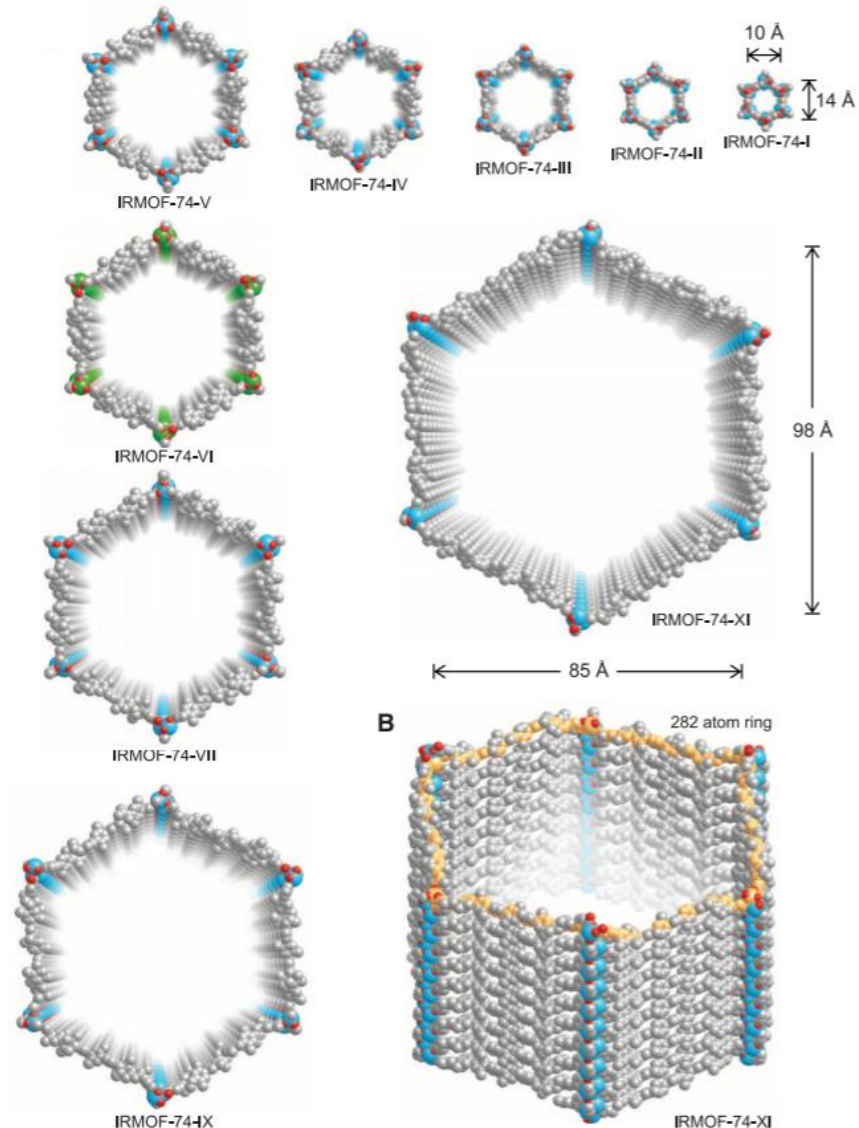
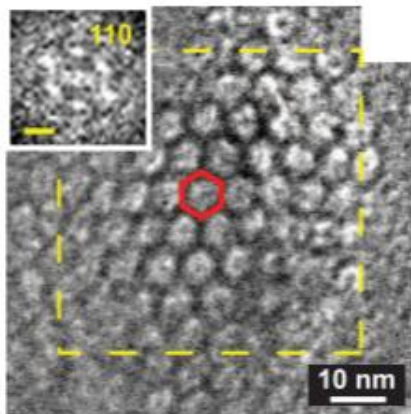
Design

(Pore opening)

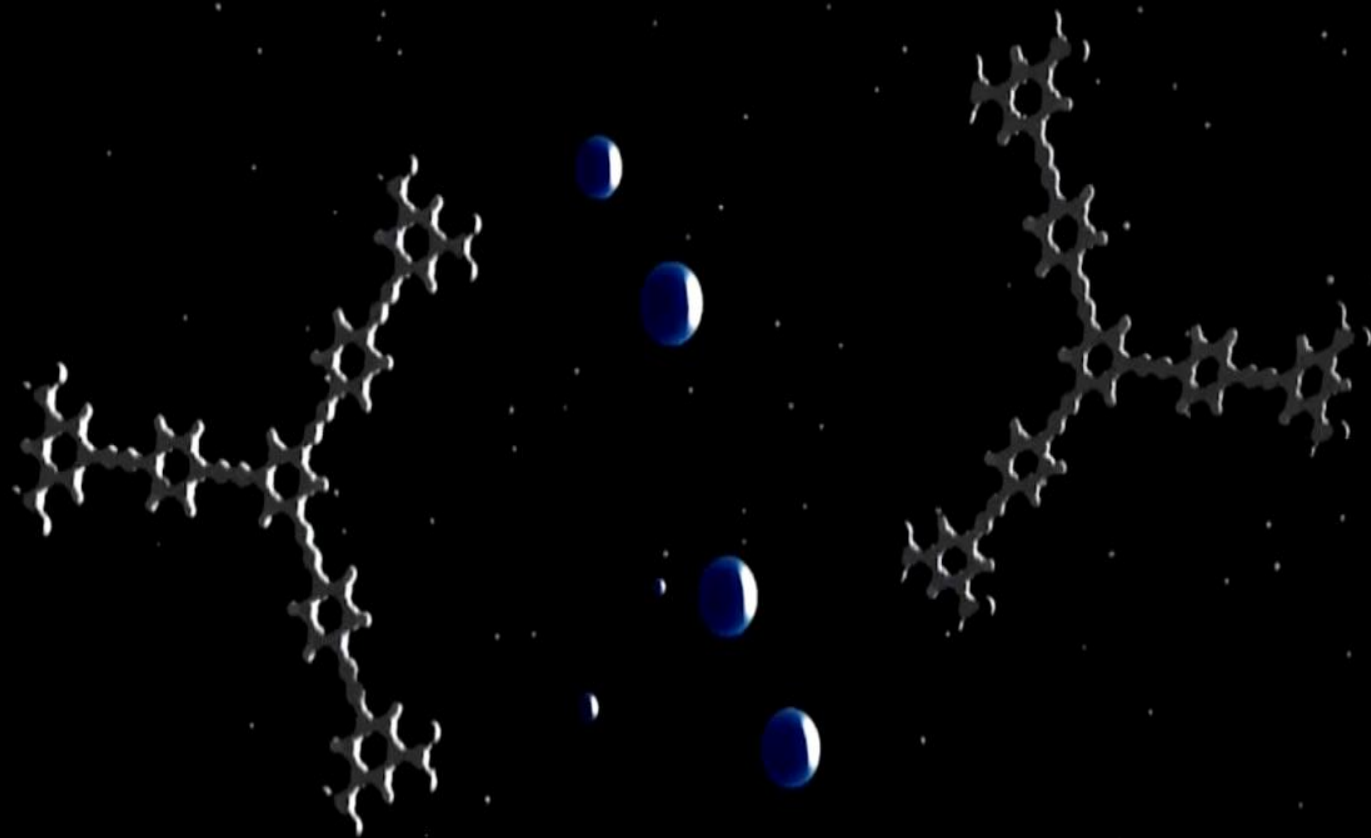
IRMOF-74-VII



IRMOF-74-IX



H. Deng, S. Grunder, K. E. Cordova, C. Valente, H. Furukawa, M. Hmadeh, F. Gándara, A. C. Whalley, Z. Liu, S. Asahina, H. Kazumori, M. O'Keeffe, O. Terasaki, J. F. Stoddart and O. M. Yaghi, *Science*, **2012**, 336(6084).



Synthesis-Activation of MOFs



As synthesized sample

- Starting material and side product in the solvent
- Starting material and side product in the pores

First solvent exchange



Solvent exchanged sample

- Solvent exchanged with solvent used in the synthesis
- Starting material and side product removed from the solvent
- Starting material and side product in the pores

Multiple cycles of solvent exchange



Fully solvent exchanged sample

- Solvent exchanged with solvent used in the synthesis
- All remaining starting materials and side products removed from the pores

Solvent exchange by low boiling point solvent



Sample ready for activation

- Solvent exchanged with low boiling point solvent
- All remaining starting materials and side products removed from the pores

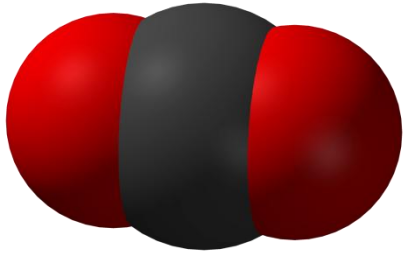


Synthesis-Activation of MOFs

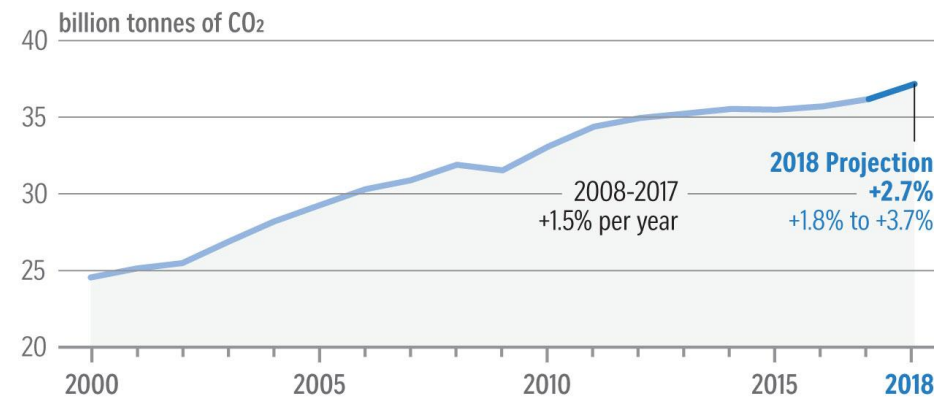
(Modulated synthesis)



Applications of MOFs



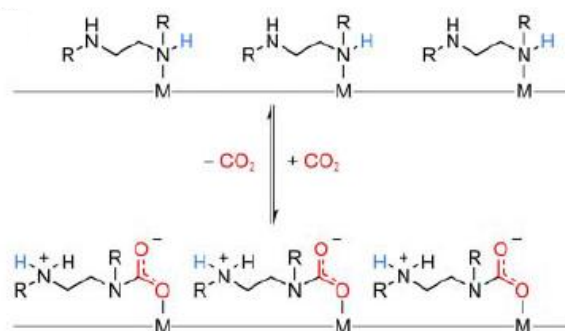
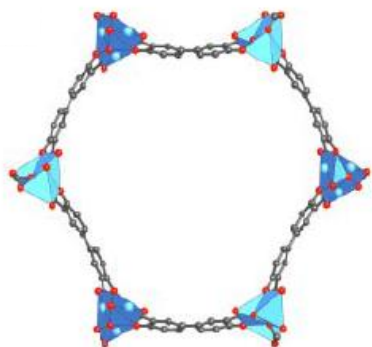
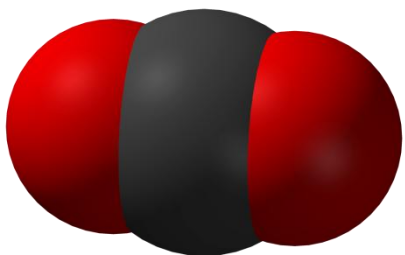
Fossil CO₂ emissions



Source: FUTURE EARTH STRAITS TIMES GRAPHICS



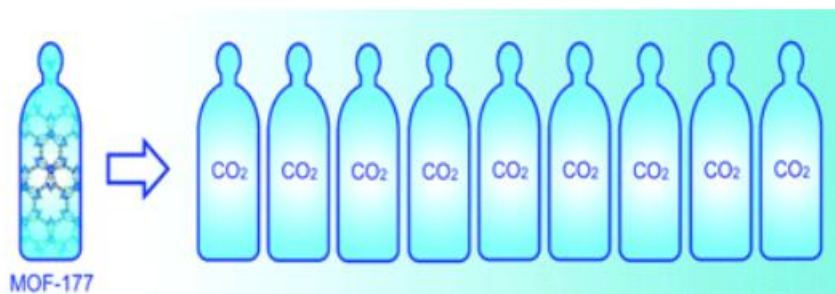
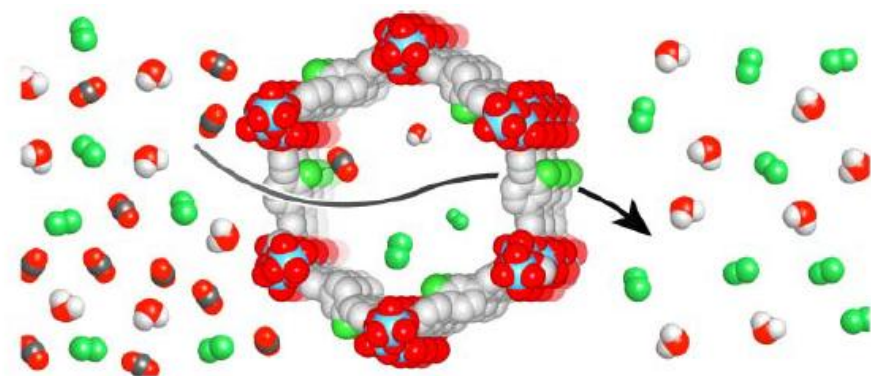
Applications of MOFs



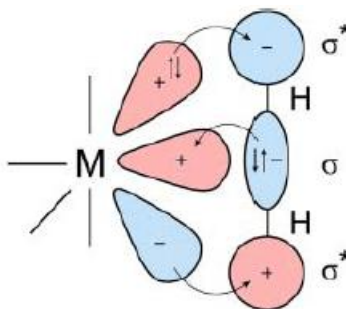
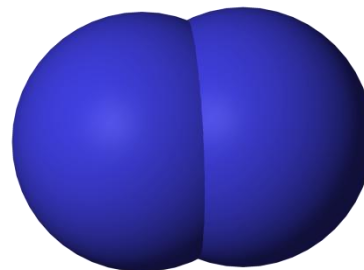
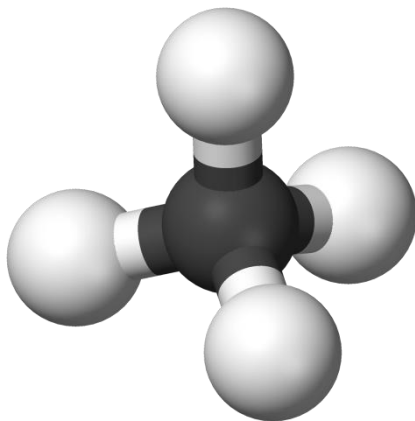
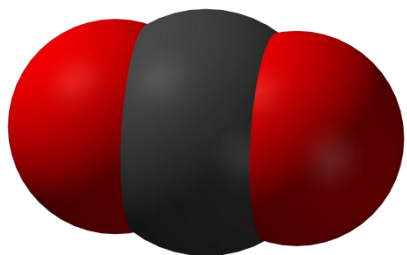
CO₂ capture/separation

Open metal sites

Amine groups

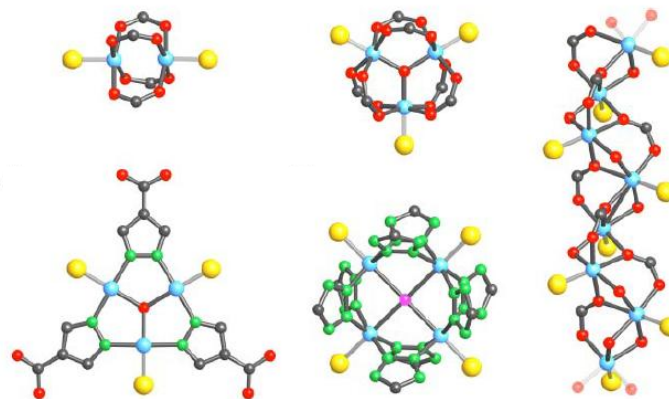
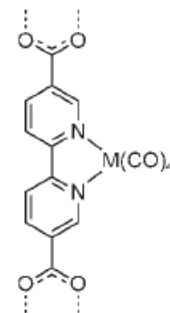
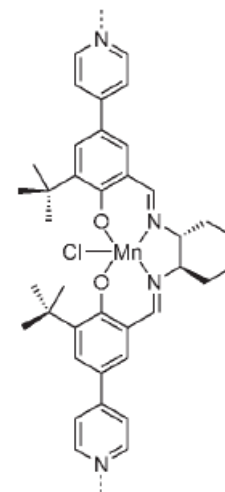


Applications of MOFs

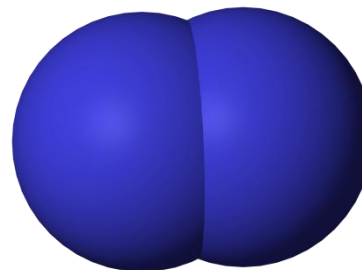
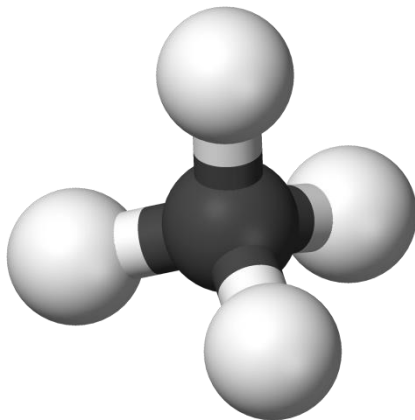
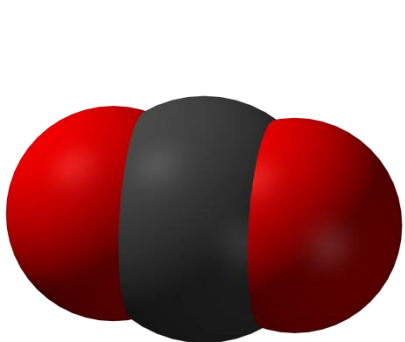


H₂ and CH₄ storage

Open metal sites



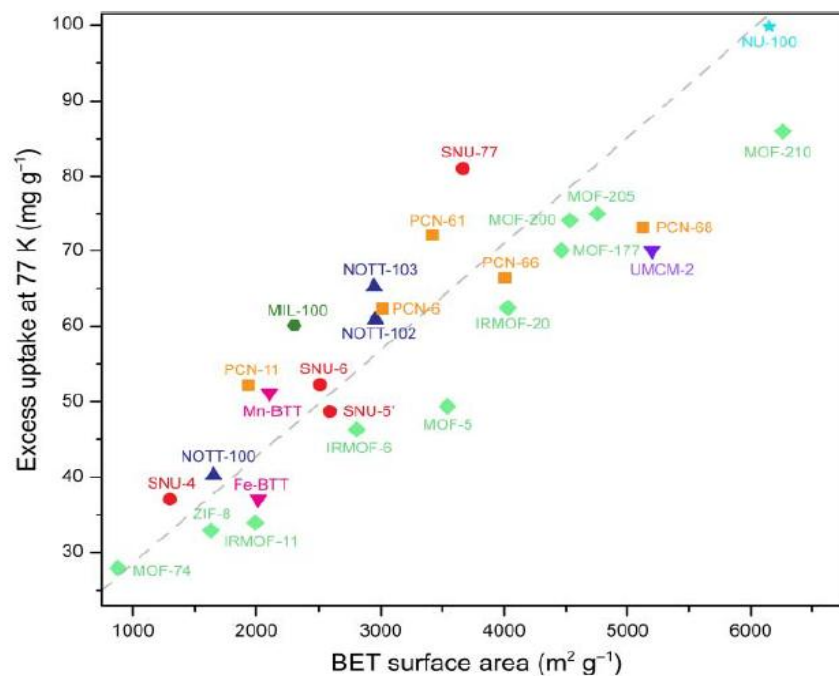
Applications of MOFs



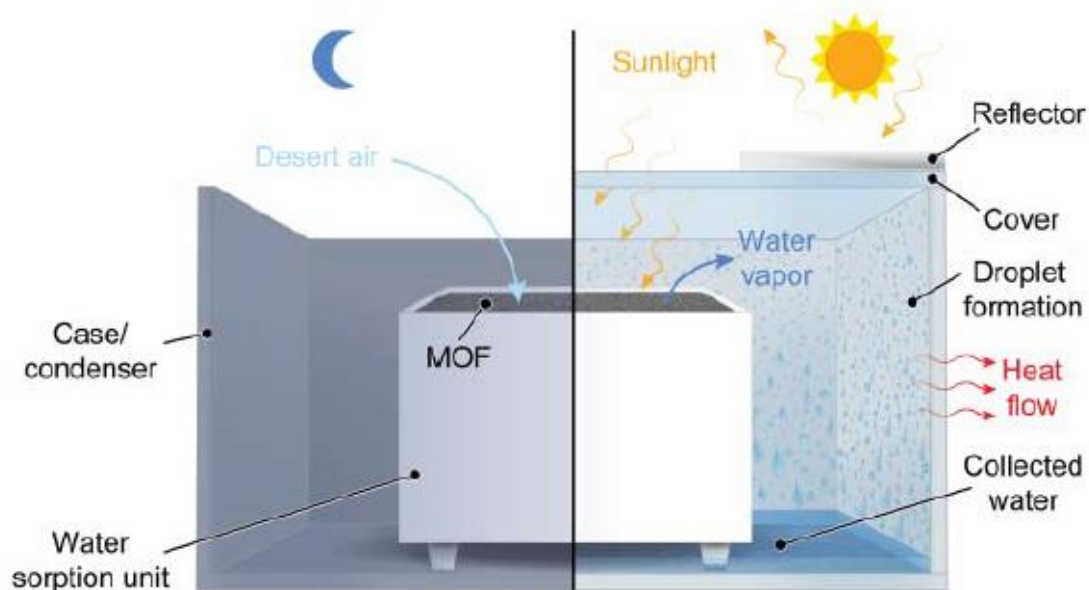
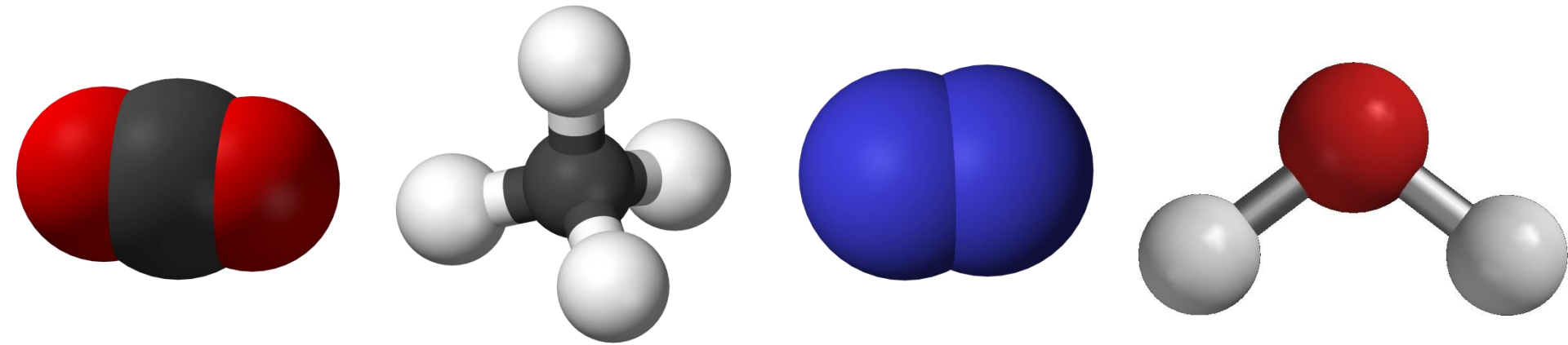
H₂ and CH₄
storage

→ Open metal sites

→ High surface areas



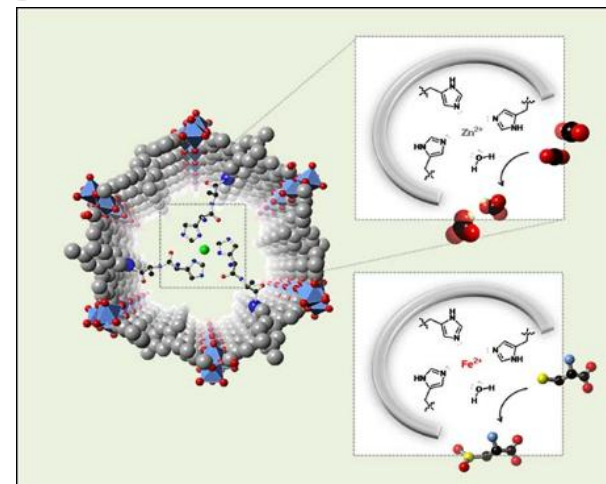
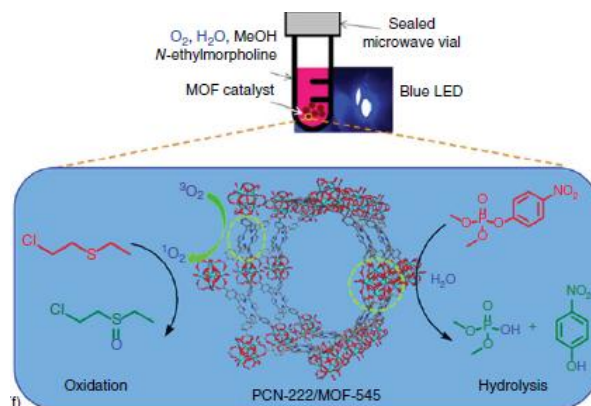
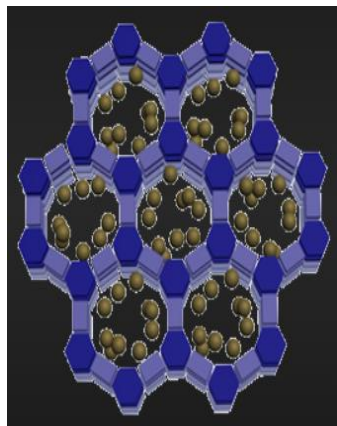
Applications of MOFs



F.Fathieh, M. J. Kalmutzki, E. A. Kapustin, P. J. Waller, J. Yang and O. M. Yaghi, *Science Advances*, **2018**, 4 (6): eaat3198.

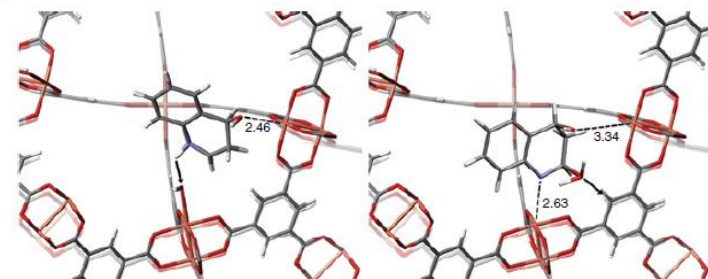
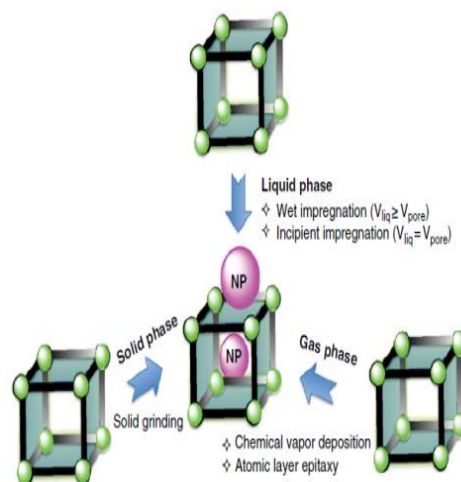
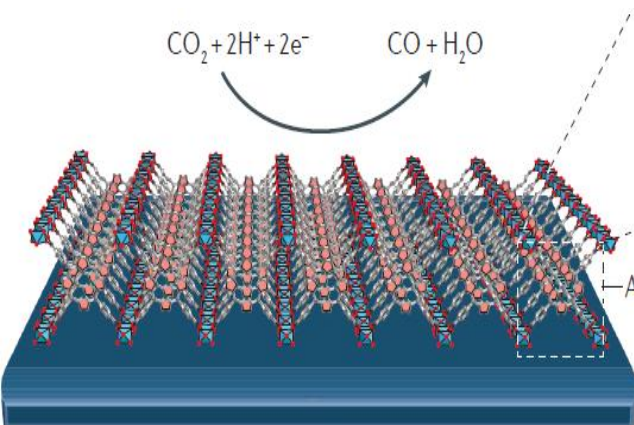


Applications of MOFs



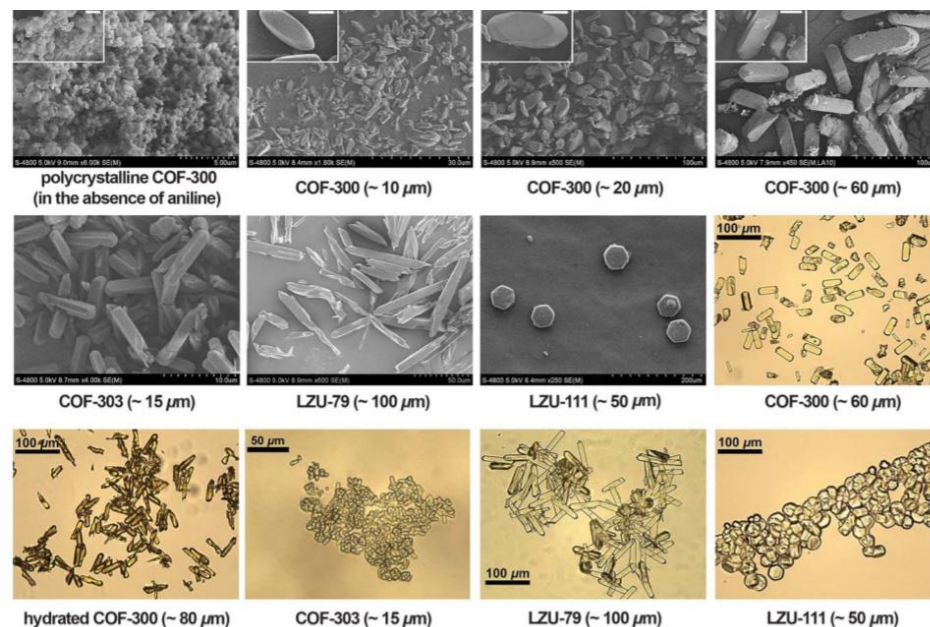
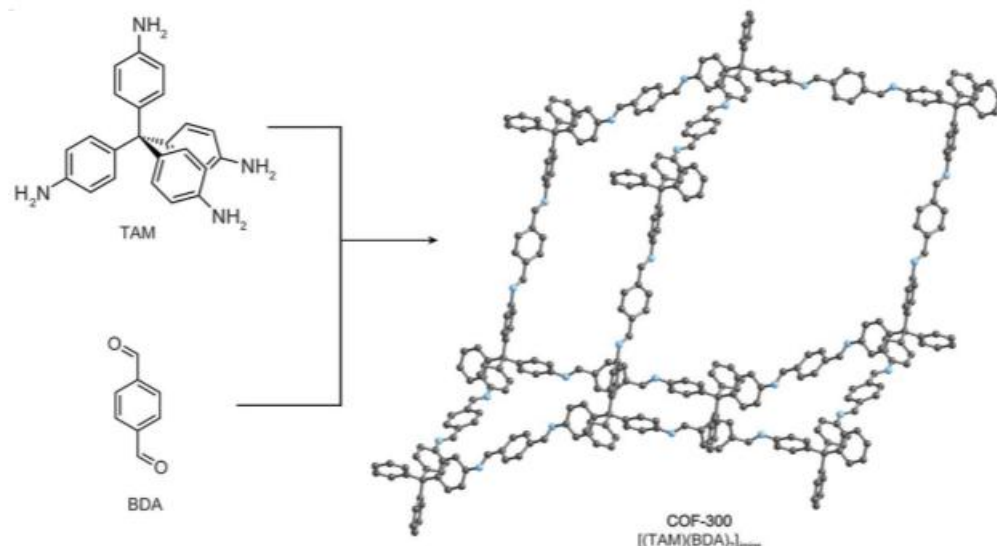
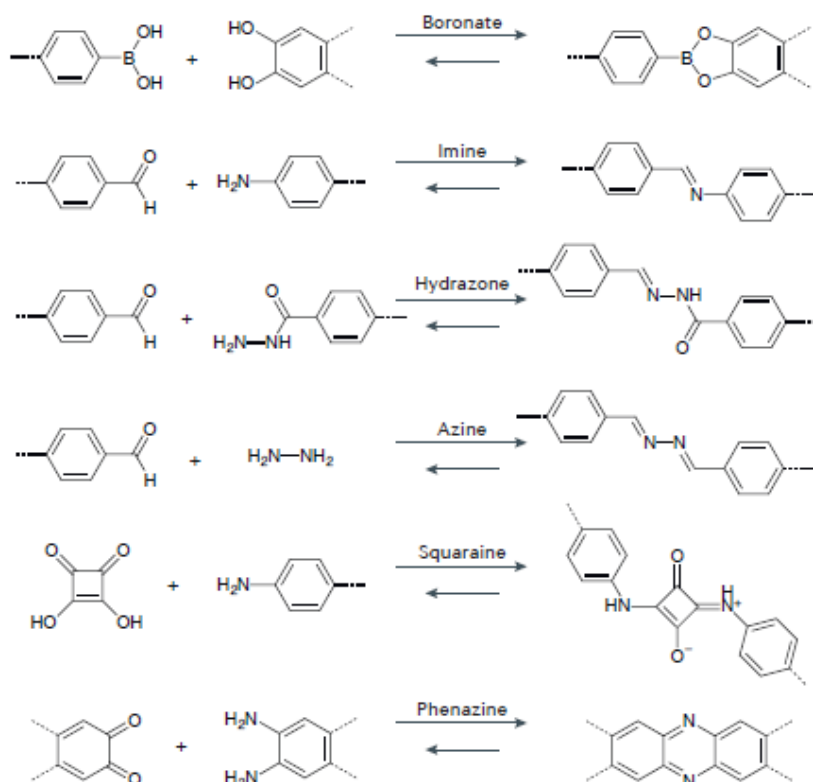
Other applications
of MOFs

- Liquid and Gas Phase separations
- Capture and Degradation of Chemical Warfare Agents
- Catalysis
- Drug Delivery
- Sensing
- Air purification



What's next?

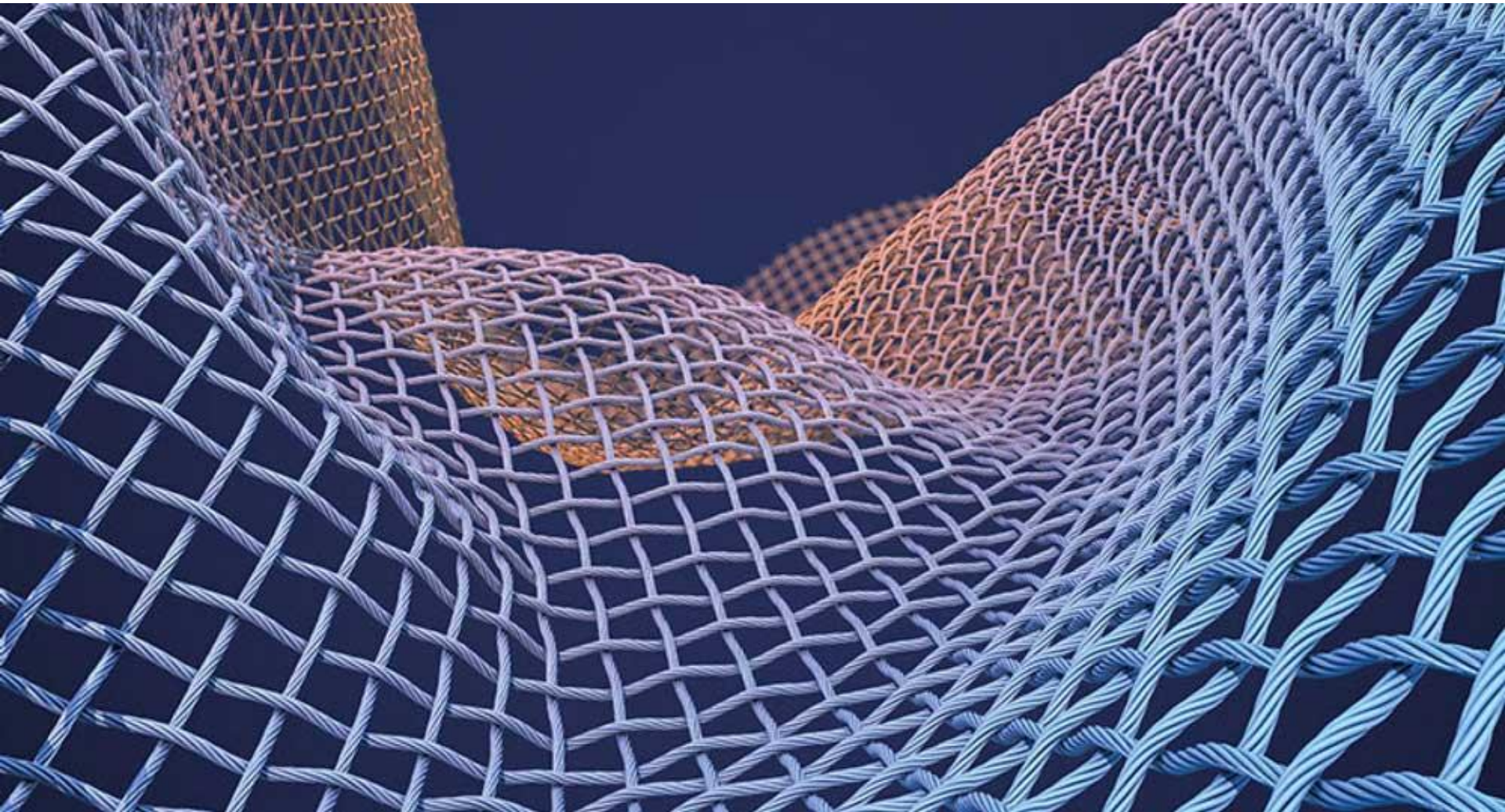
Covalent Organic Frameworks (COFs)



T. Ma, E. A. Kapustin, S. X. Yin, L. Liang, Z. Zhou, J. Niu, L. H. Li, Y. Wang, J. Su, J. Li, X. Wang, W. D. Wang, W. Wang, J. Sun and O. M. Yaghi, *Science*, **2018**, 361, 6397, 48-52.

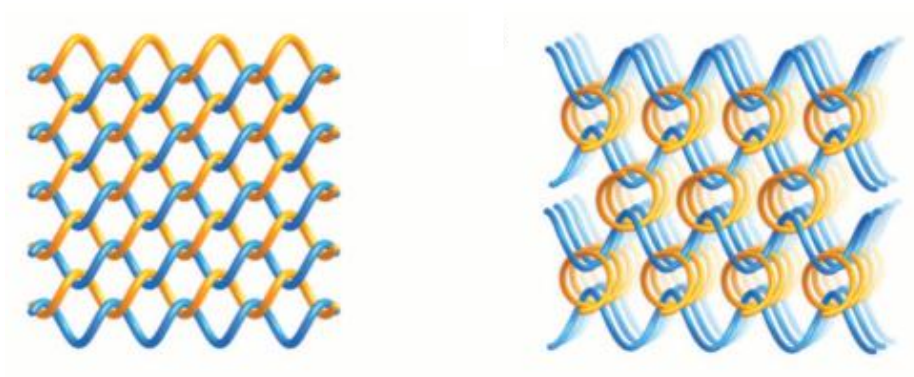
What's next?

Molecular Weaving

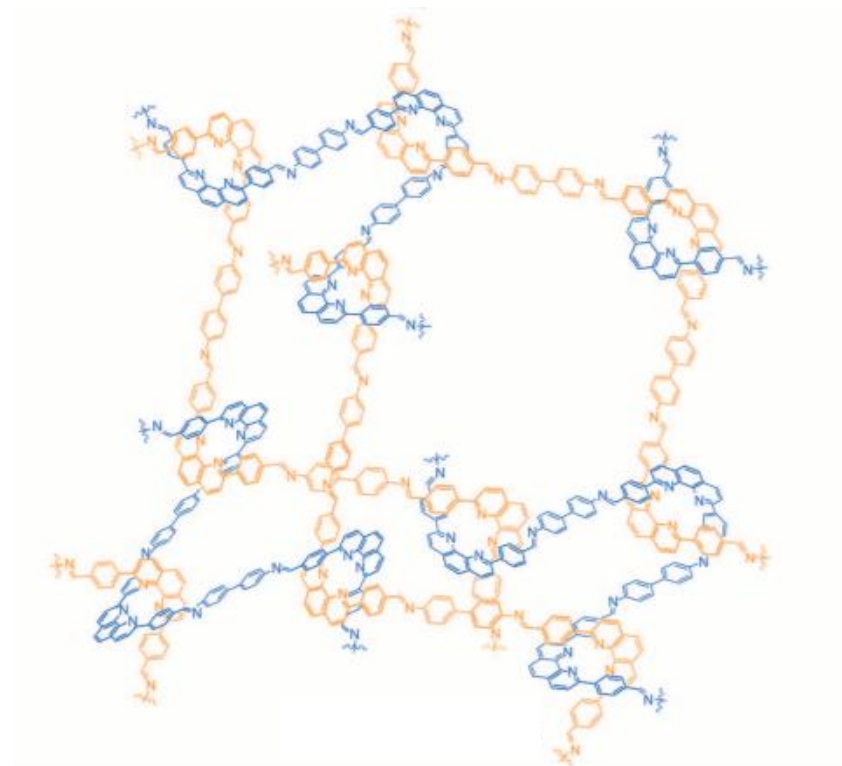


What's next?

Covalent Organic Frameworks (COFs)



Molecular Weaving



Y. Liu, Y. Ma, Y. Zhao, X. Sun, F. Gándara, H. Furukawa, Z. Liu, H. Zhu, C. Zhu, K. Suenaga, P. Oleynikov, A. S. Alshammari, X. Zhang, O. Terasaki and O. M. Yaghi, *Science*, **2016**, 365-367.

